


# Können mikrobiologische Untersuchungen helfen das Bioremediations-Potential von Chlorethen-kontaminierten Standorten abzuschätzen ?

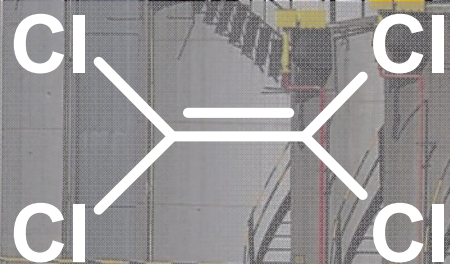


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Central Environmental Molecular Biology  
Laboratory (CEMBL)

# Chlorinated ethenes

safe solvents for industry



PCE

tetrachloroethene

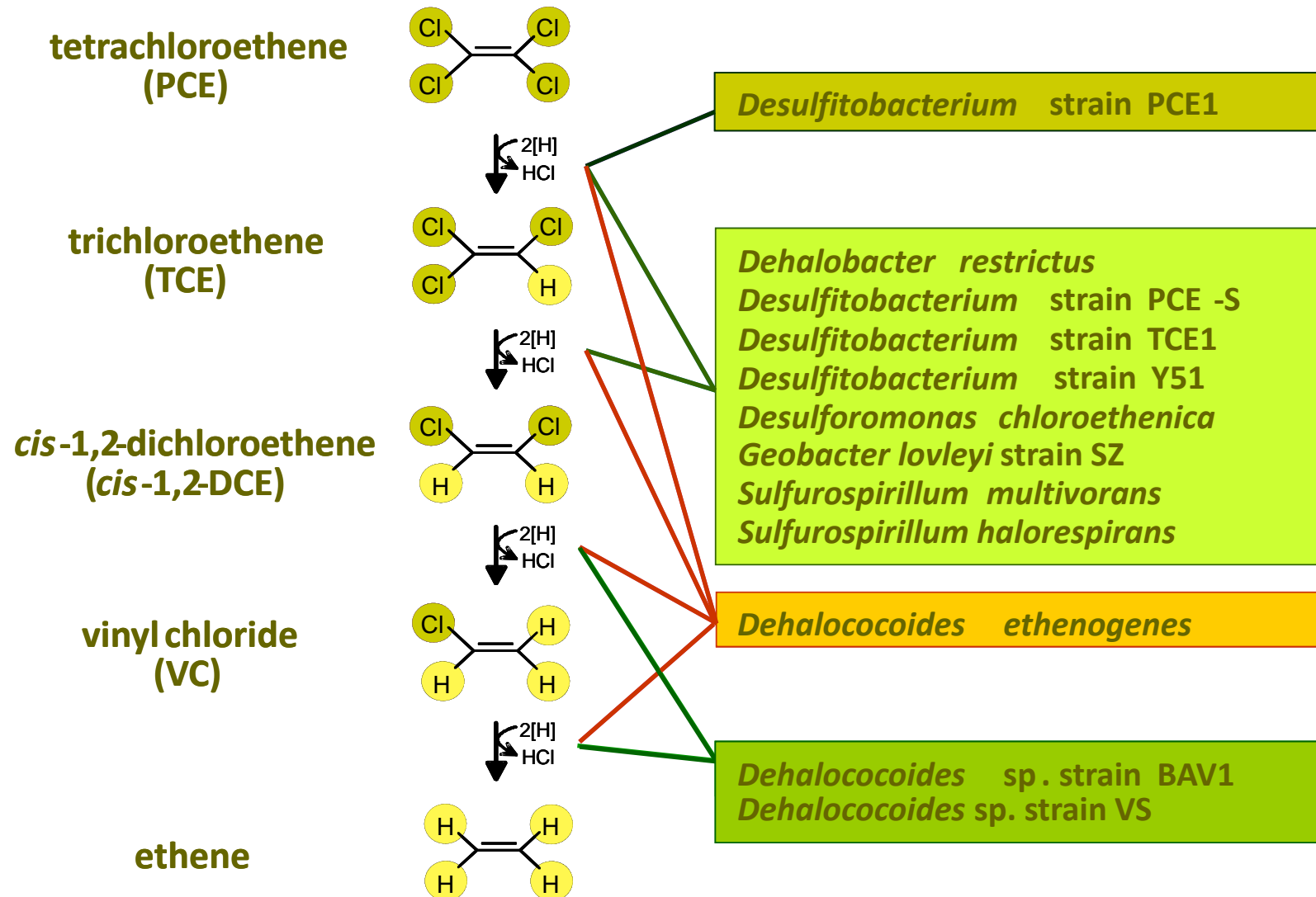


TCE

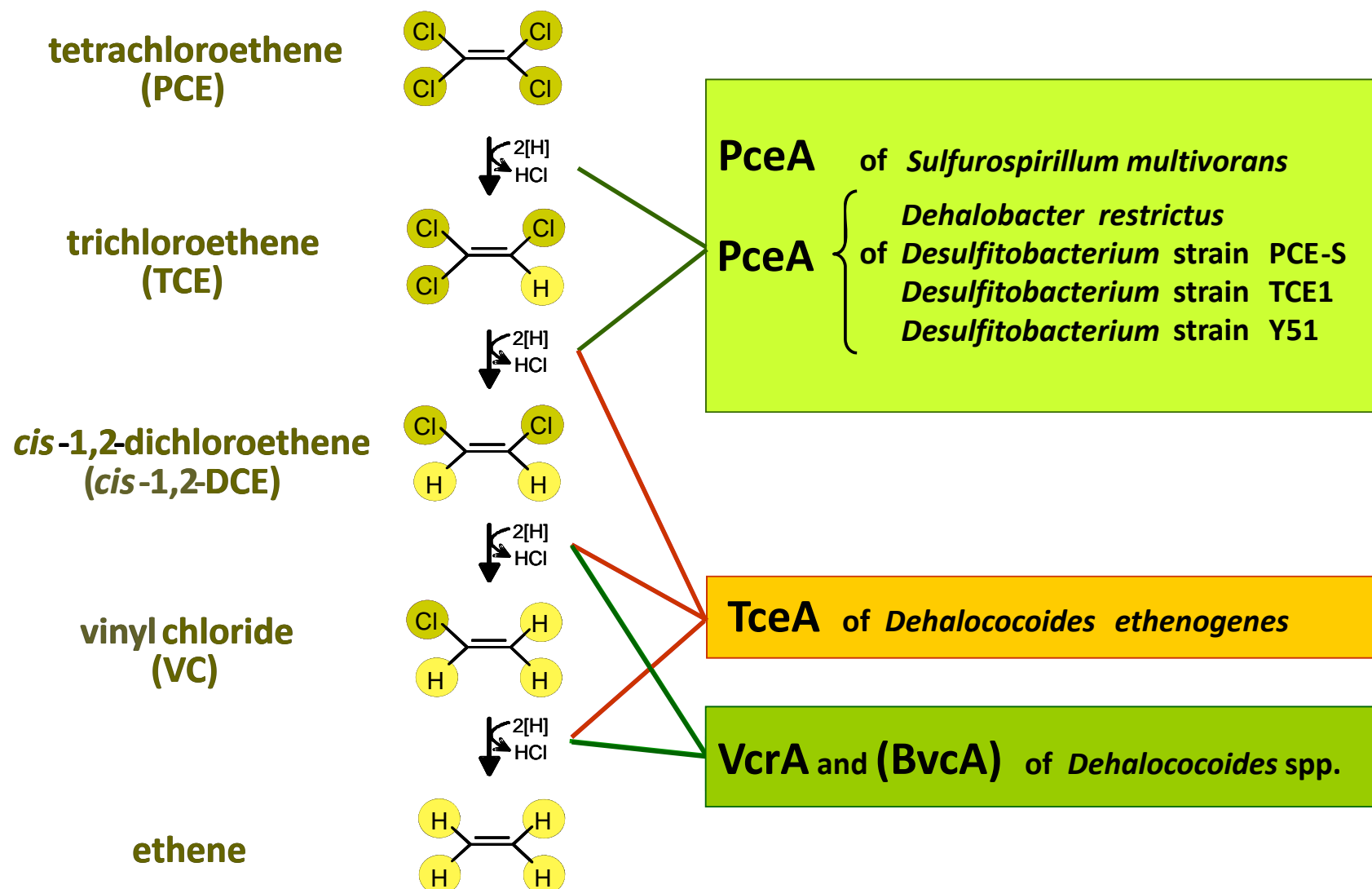
trichloroethene

recalcitrant to oxidative attack

# Isolates of chloroethene-respiring bacteria



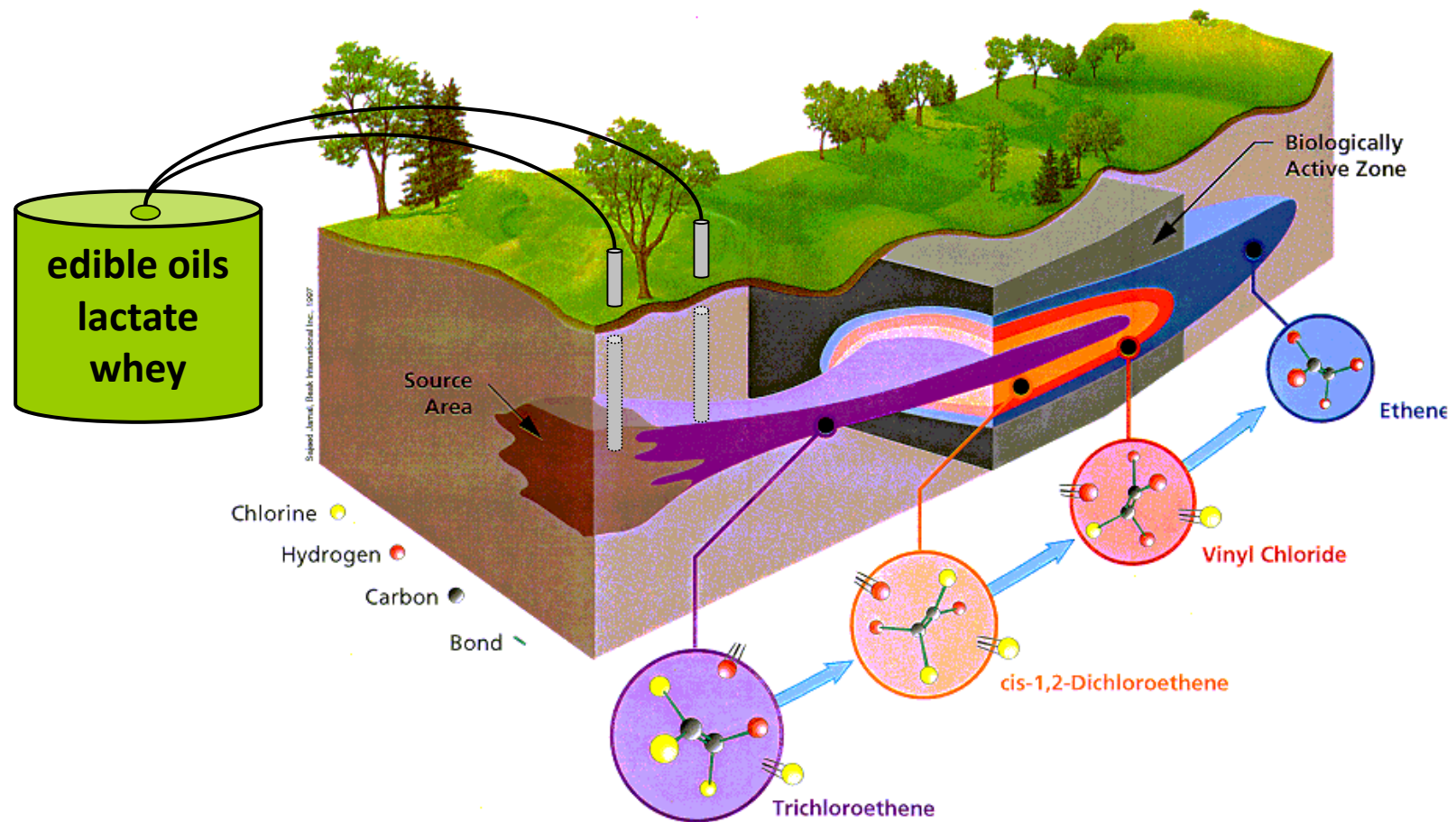
# Chloroethene reductive dehalogenases





# Their application in remediation

(enhanced) natural attenuation

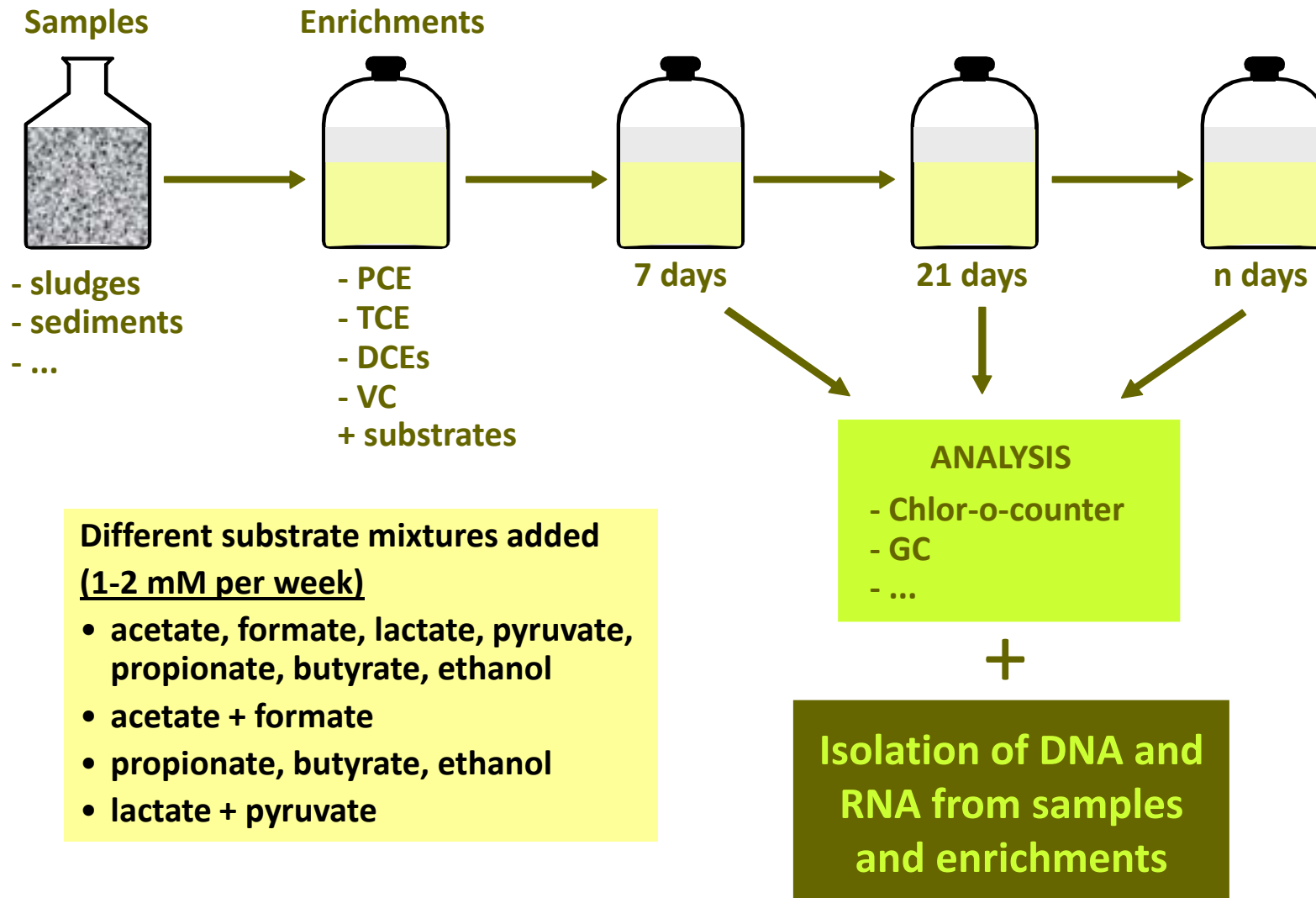


# Challenges for (enhanced) natural attenuation of chloroethene contamination

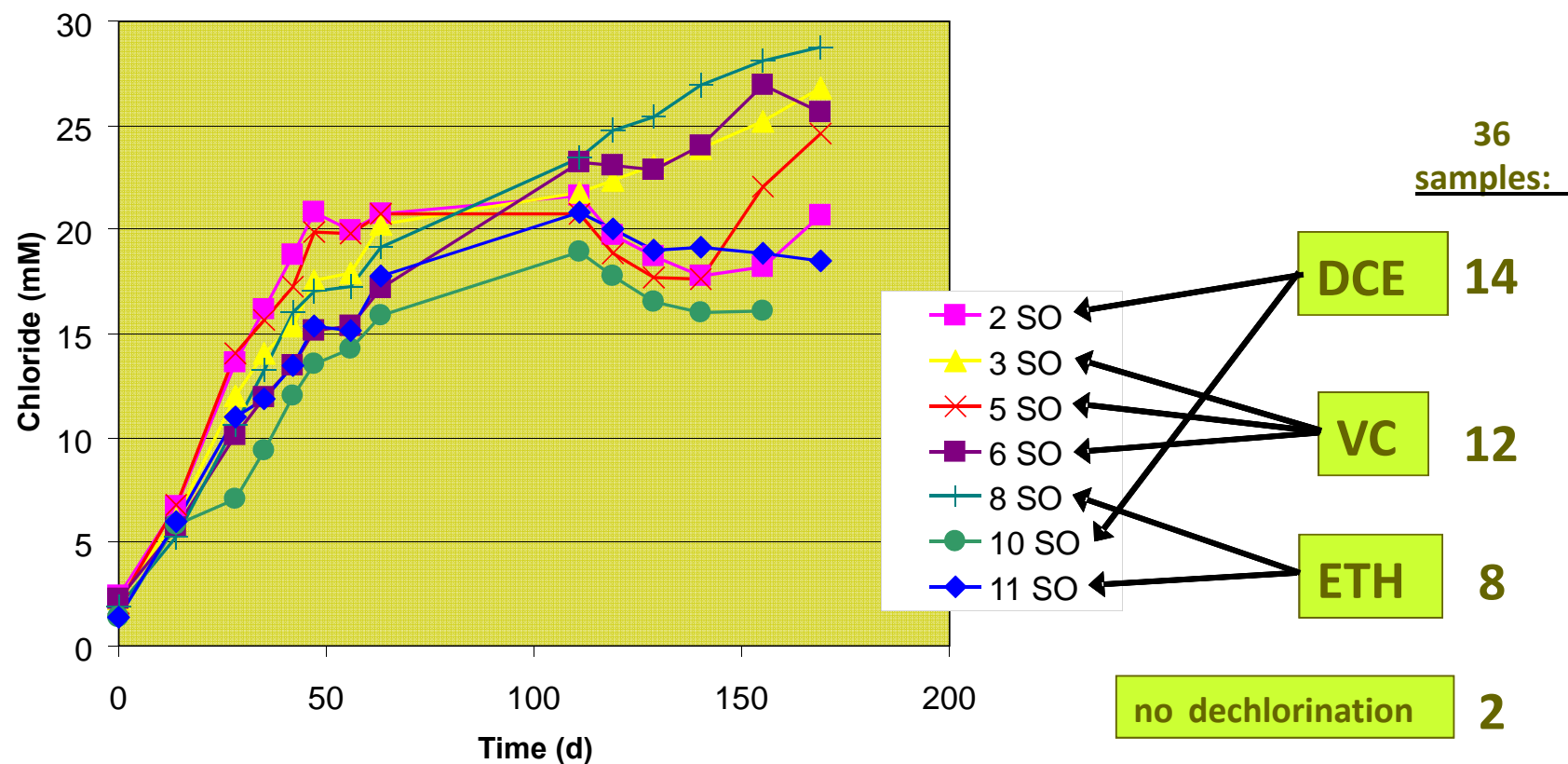
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- Assure that vinyl chloride is not end product
    - Is the biochemical potential present for complete chloroethene dechlorination ?
    - If yes, why does vinyl chloride accumulate ?
    - For enhanced NA which electron donors sustain best complete dechlorination ?
  - Understand the fate of toxic dechlorination products
    - Is dechlorination the only transformation process responsible for VC removal ?
-

# Experimental set-up of microcosms

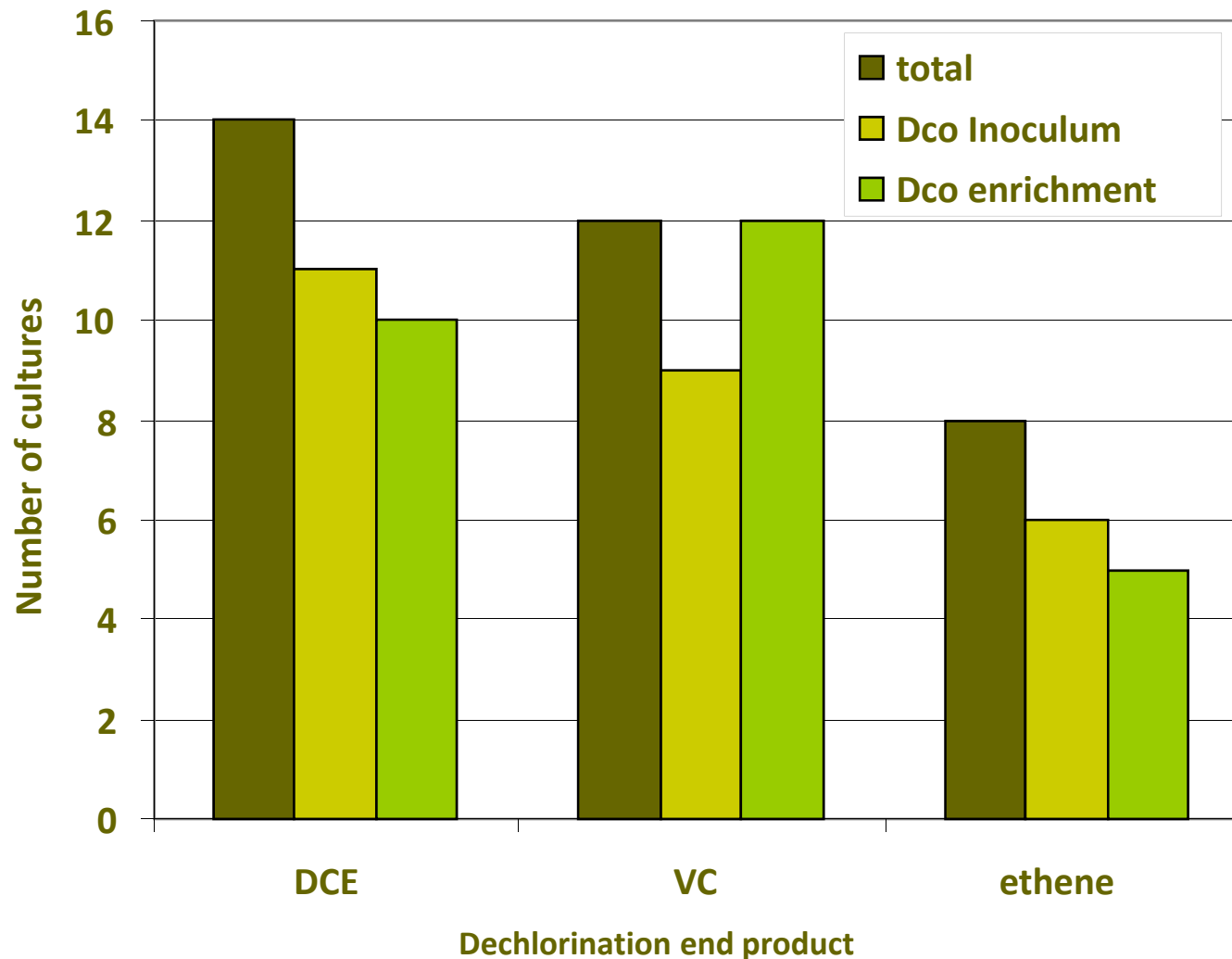


# Dechlorination in enrichment cultures

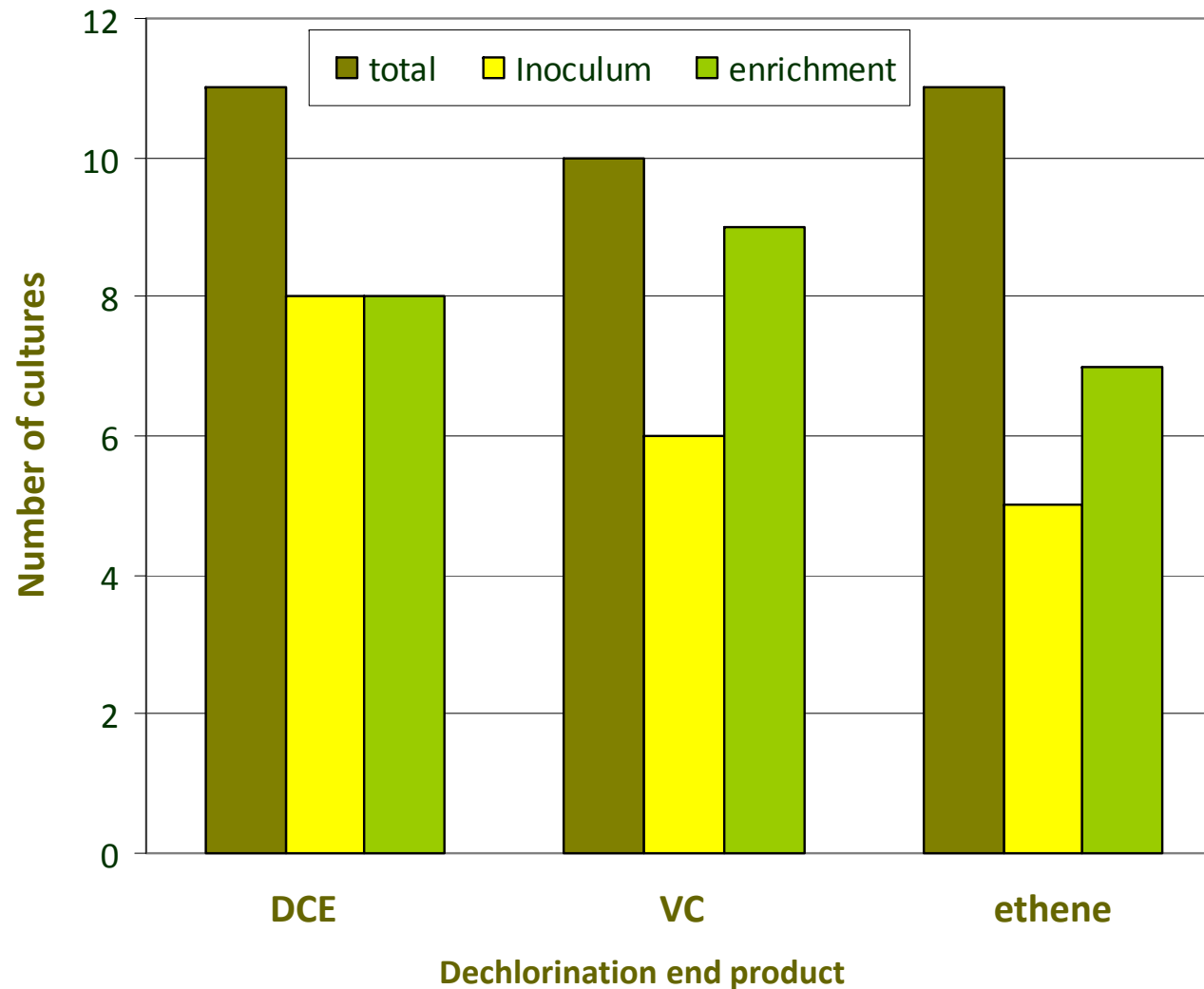




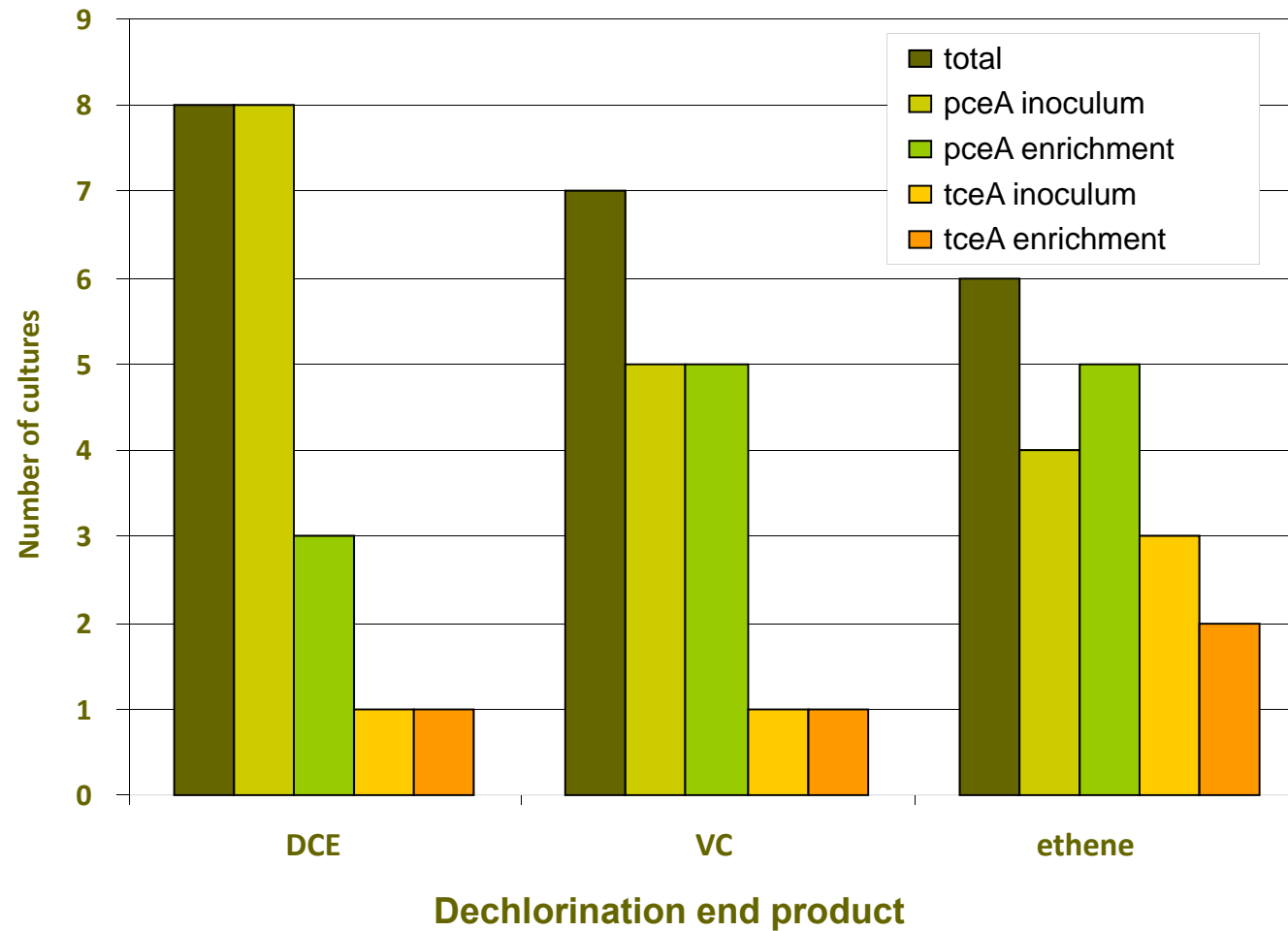
# Detection of *Dehalococcoides* spp. in 36 environmental samples



# Detection of *Dehalococcoides* spp. together with *Desulfitobacterium* spp. in 24/32 samples



# Detection of *pceA* et *tceA* in environmental samples



Inoculum: nested PCR; enrichment: direct PCR

# Stepwise dechlorination by different organohalide-respiring bacteria

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## □ PCE → DCE → ethene:

- *Desulfitobacterium* spp. → *Dehalococcoides* spp.

## □ PCE → DCE → ethene:

- *Sulfurospirillum* spp. → *Dehalococcoides* spp.

## □ PCE → DCE → ethene:

- *Dehalobacter* spp. → *Dehalococcoides* spp.

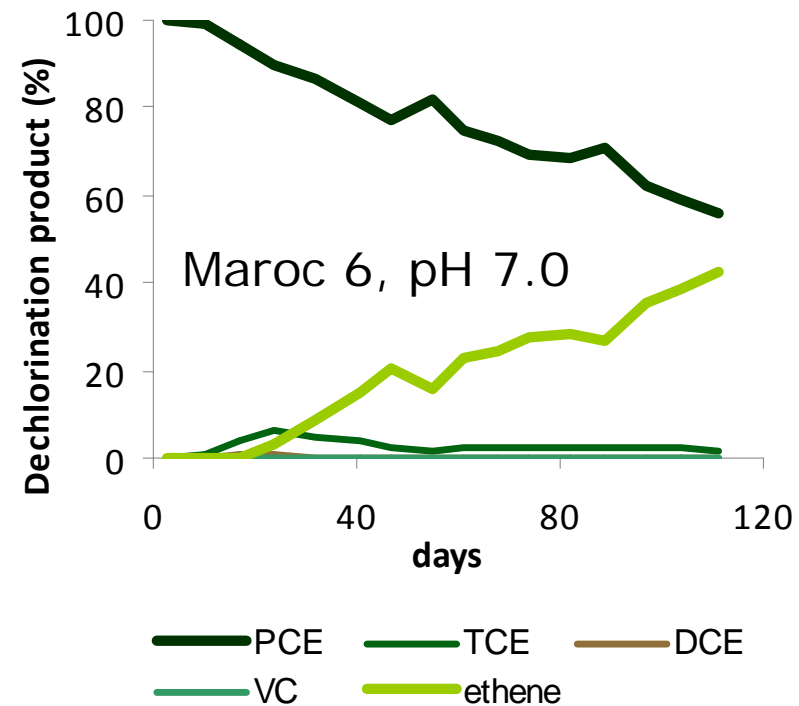
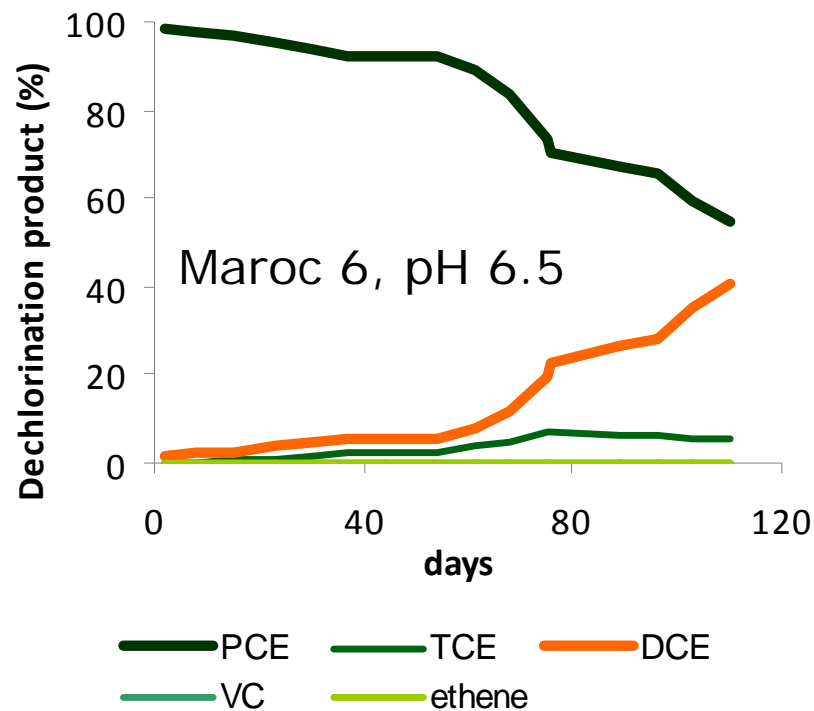
- **no vcrA and bvcA detected**

## □ DCE → VC → ethene:

- *Desulfitobacterium* spp. → *Dehalococcoides* spp.

- **no vcrA and bvcA detected**
-

# The *Dehalobacter-Dehalococcoides* culture



- ❑ very slow dechlorination
- ❑ production of DCE with low accumulation of TCE at pH 6.5
- ❑ production of ethene with low accumulation of TCE and no accumulation of DCE and VC at pH 7.0

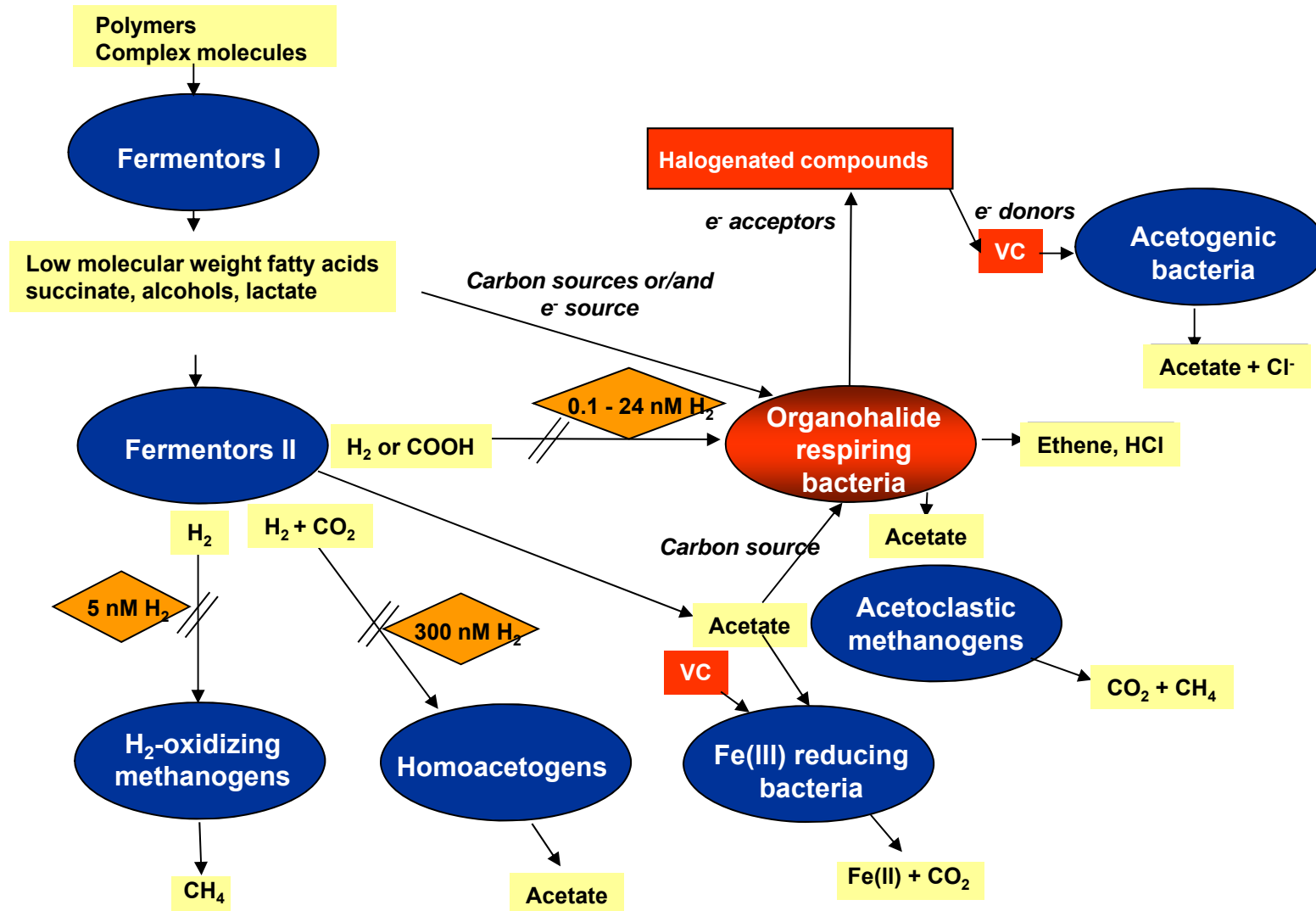


# Challenges for (enhanced) natural attenuation of chloroethene contamination

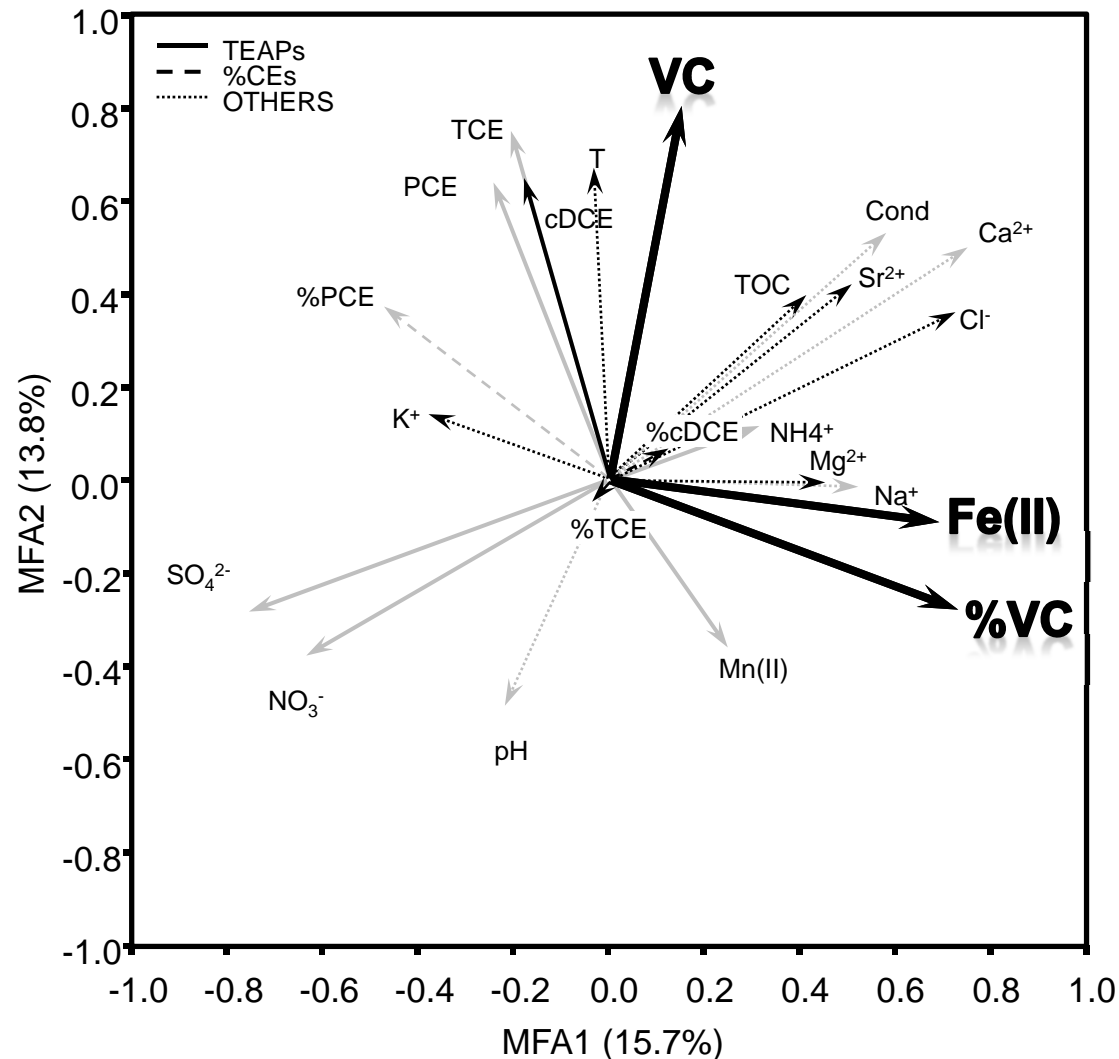
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# Bacterial guilds involved in organohalide respiration



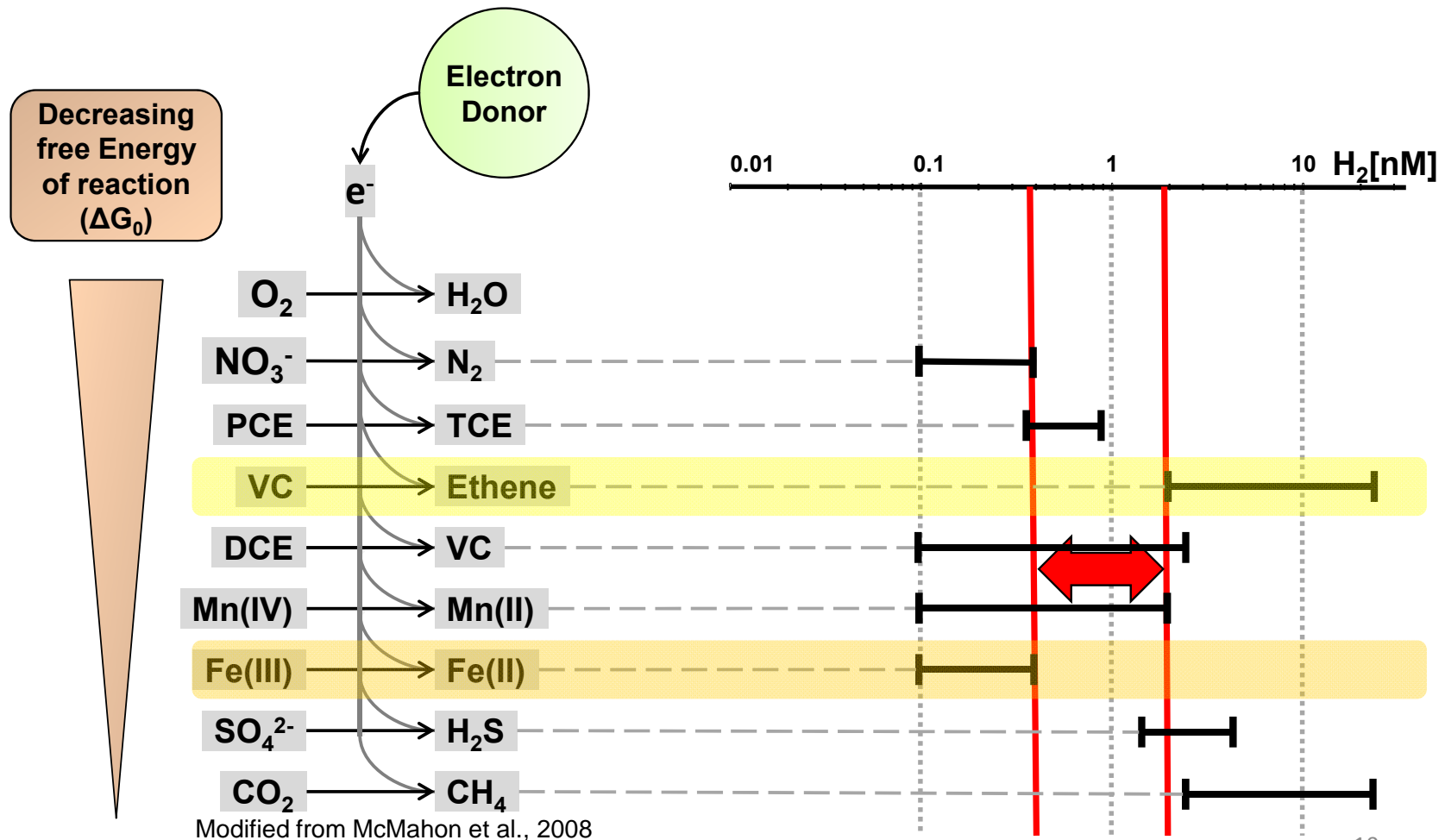
# Competition between iron-respiring and organohalide-respiring bacteria



# Competition between iron-respiring and organohalide-respiring bacteria

Ecological succession of electron acceptors

Required  $H_2$  concentrations



# Challenges for (enhanced) natural attenuation of chloroethene contamination

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## Electron donors for complete dechlorination

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Inoculum	Electron donor mixture leading to ethene formation
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Anaerobic digester sludge	Acetate - formate
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PCE bioreactor sludge	Propionate – butyrate - ethanol
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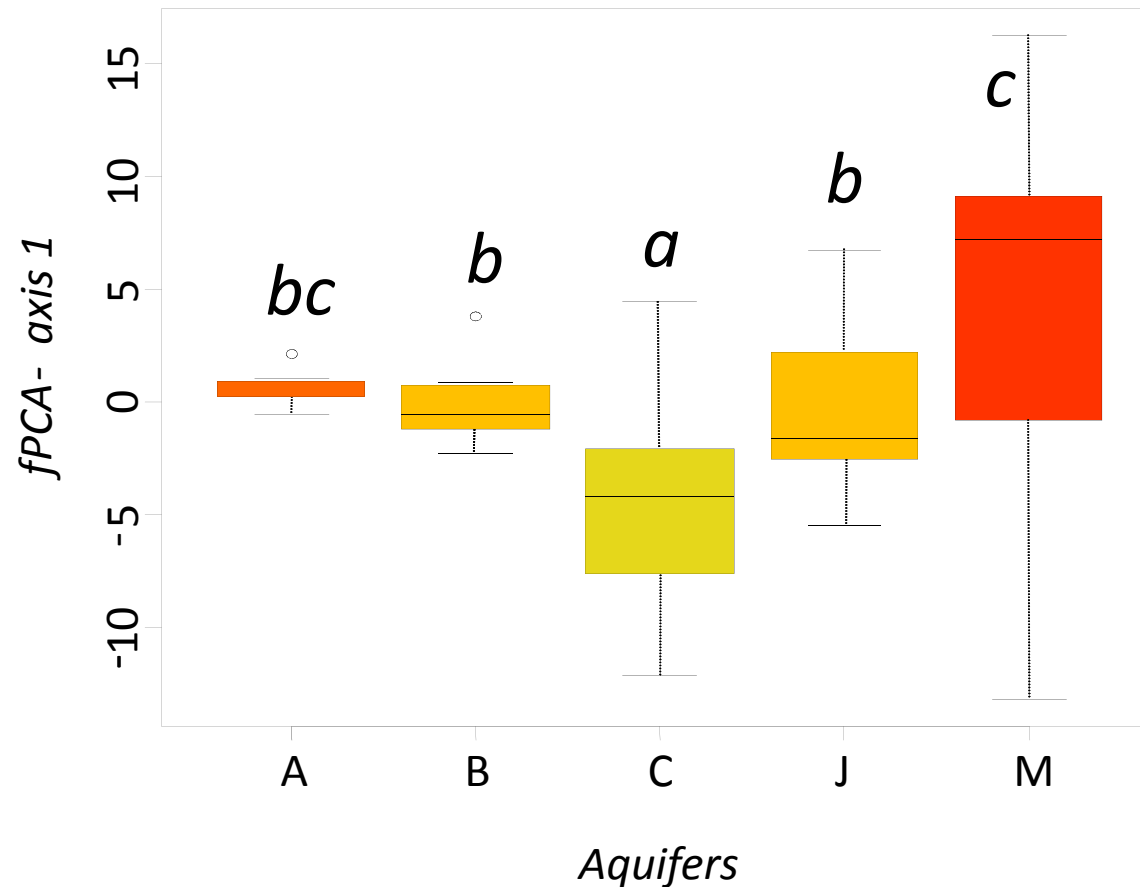
Aquifer I	Propionate – butyrate - ethanol
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Aquifer II	Propionate – butyrate - ethanol
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# Bacterial diversity and microbial community composition varies according to aquifers



*Coordinates of record calculated on the first ordination axis of the fPCA, grouped by aquifers. Different letters indicate significant differences between the microbial communities of the five aquifers ( $P \leq 0.05$  Tukey's HSD).*

# Challenges for (enhanced) natural attenuation of chloroethene contamination

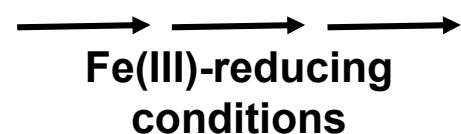
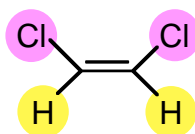
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# Anaerobic oxidation of chloroethenes

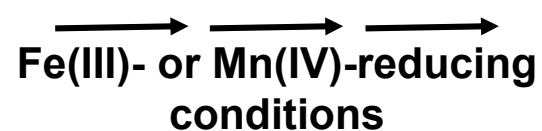
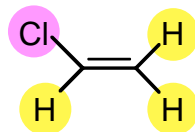
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*cis*-1,2-dichloroethene  
(*cis*-1,2-DCE)



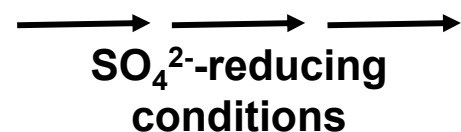
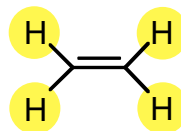
CO<sub>2</sub>

vinyl chloride  
(VC)



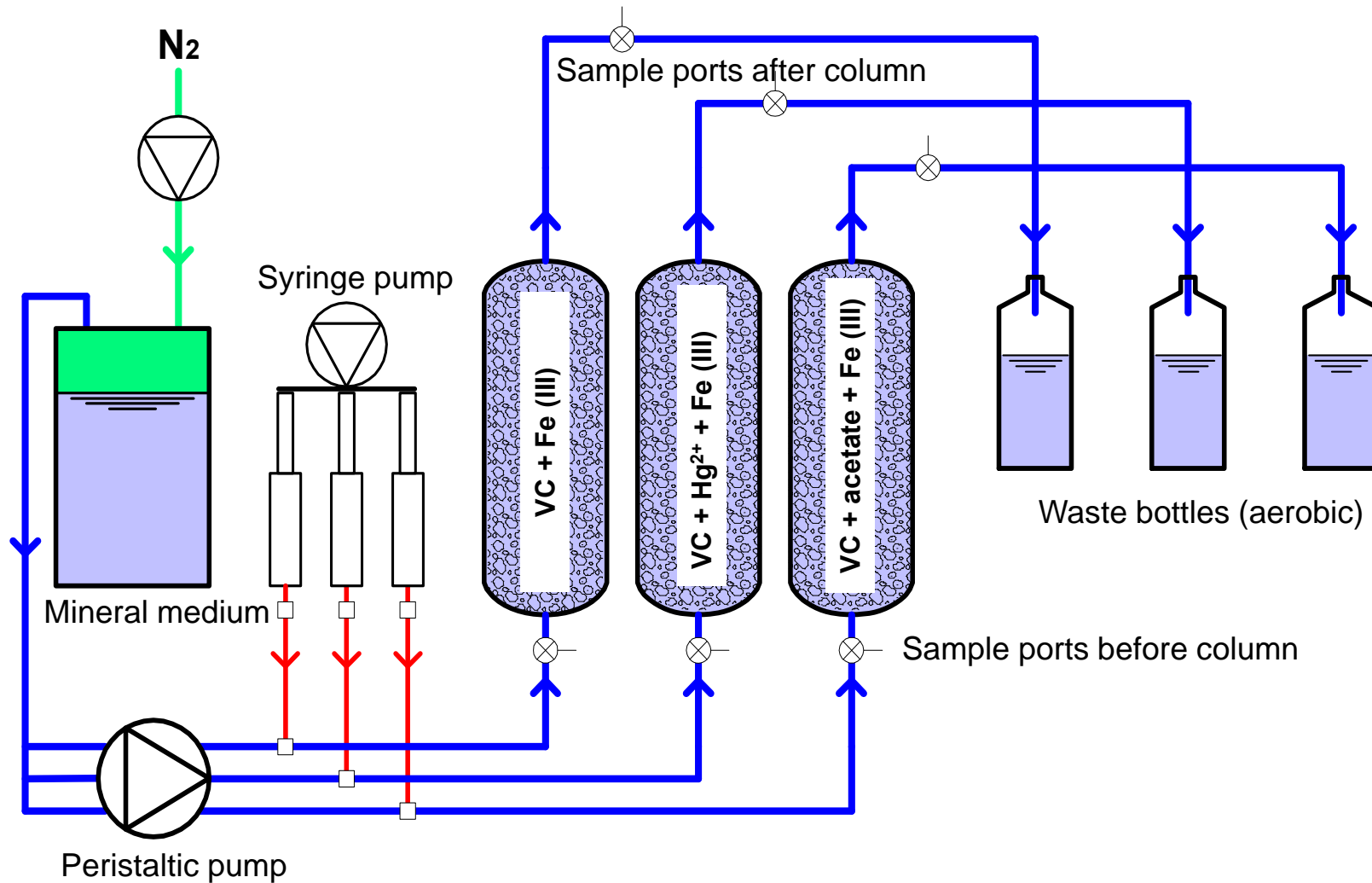
CO<sub>2</sub>

ethene



CO<sub>2</sub>

# Scheme of experimental system





# Stable isotopes indicated anaerobic oxidation of VC

$$\delta^{13}\text{C}_{\text{out}} = \delta^{13}\text{C}_{\text{in}} + \epsilon \ln(\text{C}_{\text{out}}/\text{C}_{\text{in}})$$

$\epsilon$  = isotope enrichment factor

		$\epsilon$
→	VC + Fe (III)	0-20% VC no ethene some Fe(II)  - (3 – 5)
→	VC + acetate + Fe (III)	20-50% VC ethene Fe(II)  - (11 – 18)*

\* literature data for reductive dechlorination  $\epsilon$  = - (20-25)

# Conclusions

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- Reductive dechlorination of chloroethenes is a promising microbial process for aquifer bioremediation
  - Organohalide-respiring bacteria seem to be widespread but not all biochemical potential seems to be everywhere
  - We do not know yet all the microbiology and enzymology involved in complete dechlorination to ethene
  - VC accumulation can be the result of missing biochemical potential or a consequence of losing competition for substrate
  - VC can also be transformed by processes other than reductive dechlorination
  - There is no general concept that can be applied to all sites, each site has its own characteristics and individual solutions have to be found
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# Final Conclusion

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- The bioremediation potential can be assessed with microbial investigations and a combination of molecular ecology, microcosms, and numerical ecology is the optimal approach to predict qualitatively the fate of chloroethenes during (enhanced) natural attenuation
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