

# Können mikrobiologische Untersuchungen helfen das Bioremediations-Potential von Chlorethen-kontaminierten Standorten abzuschätzen ?

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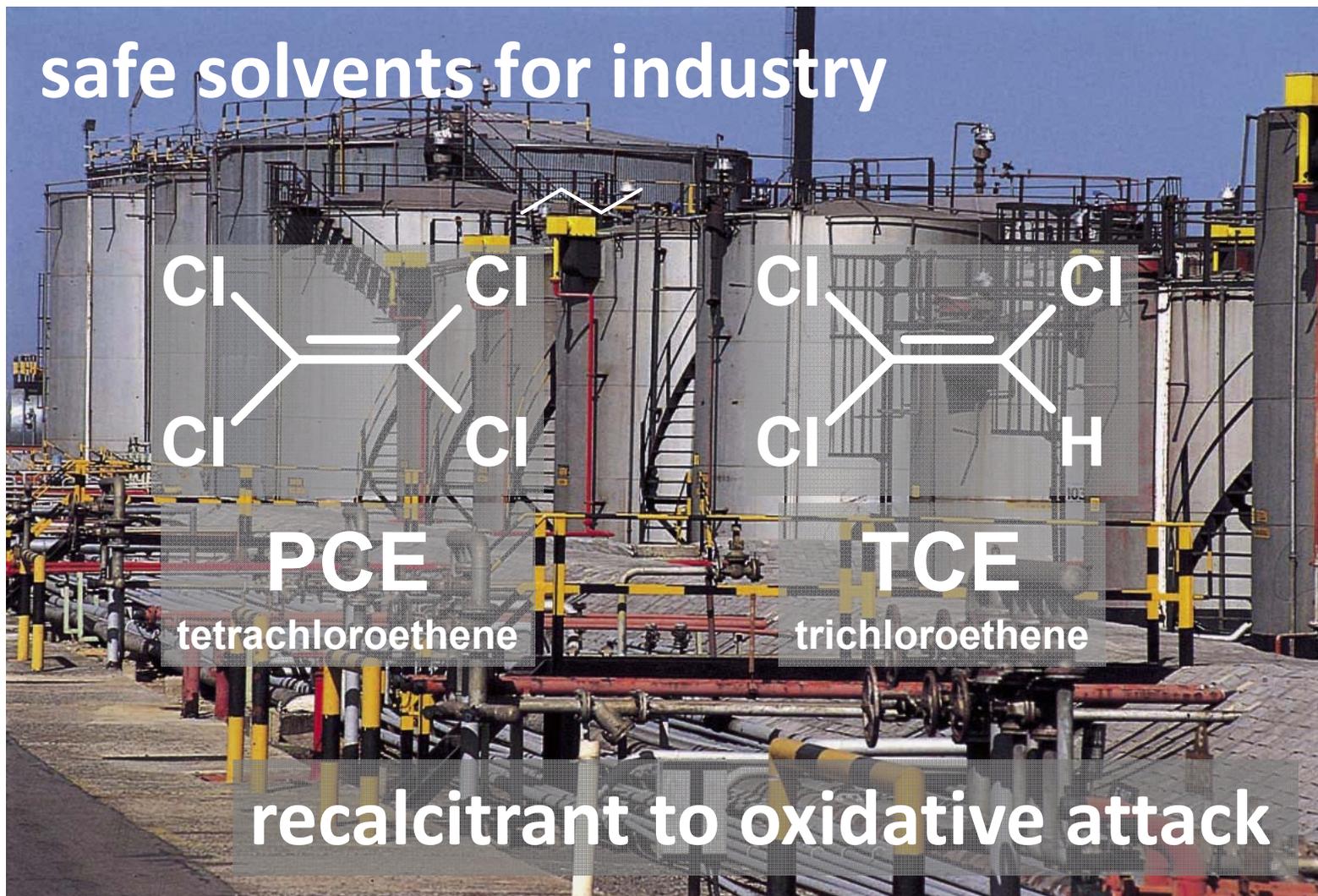
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Laboratory (CEMBL)

# Chlorinated ethenes

safe solvents for industry



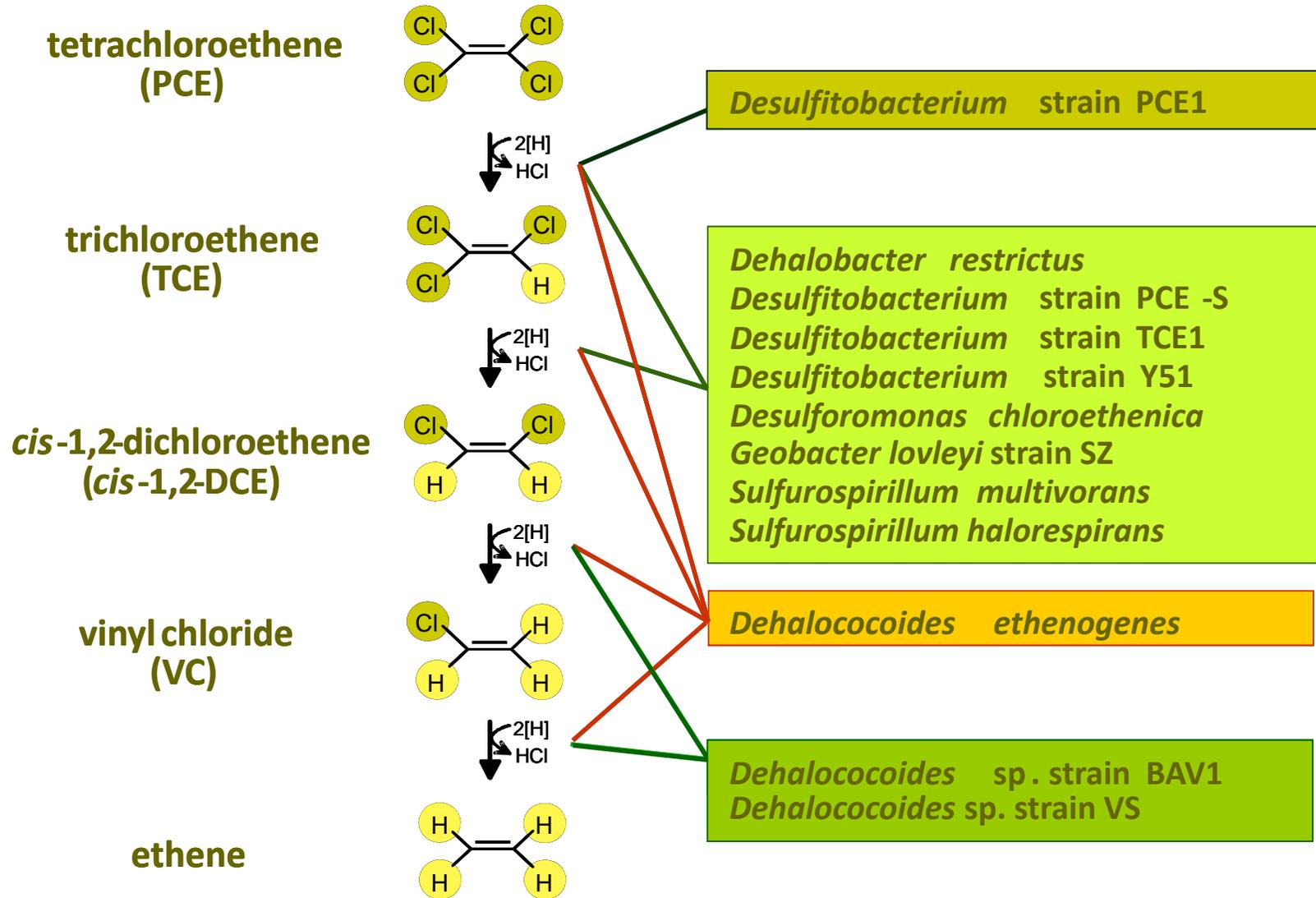
The image shows an industrial facility with several large, cylindrical storage tanks and a network of pipes and walkways. The scene is set outdoors under a clear blue sky. Overlaid on the image are two chemical structures. The structure on the left is tetrachloroethene (PCE), represented as a central carbon-carbon double bond with four chlorine atoms (Cl) attached to the carbons. The structure on the right is trichloroethene (TCE), represented as a central carbon-carbon double bond with three chlorine atoms (Cl) and one hydrogen atom (H) attached to the carbons. Below each structure are its respective abbreviations and full names: PCE (tetrachloroethene) and TCE (trichloroethene).

ClC(Cl)=C(Cl)Cl  
PCE  
tetrachloroethene

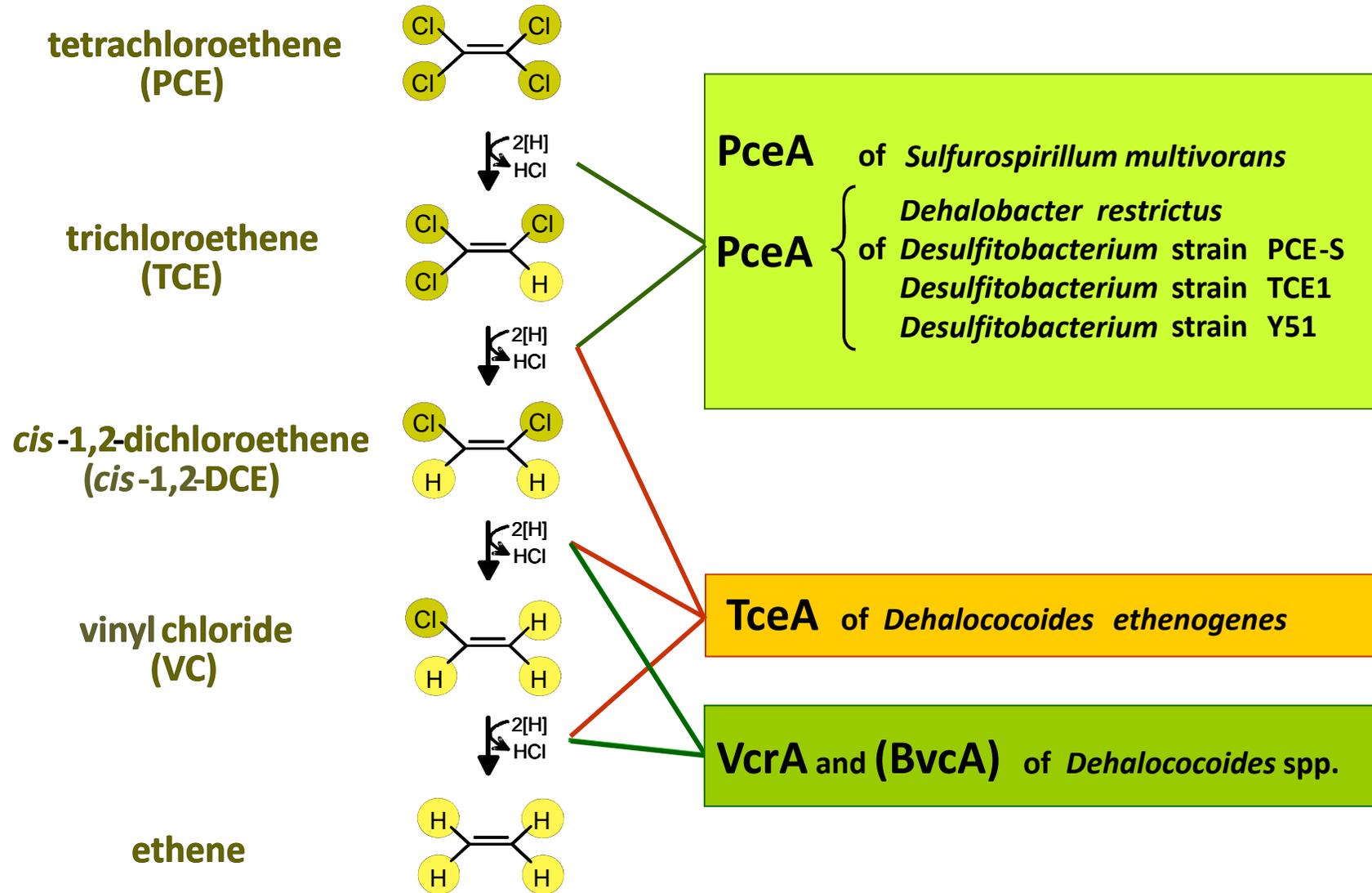
ClC(Cl)=C(Cl)H  
TCE  
trichloroethene

recalcitrant to oxidative attack

# Isolates of chloroethene-respiring bacteria

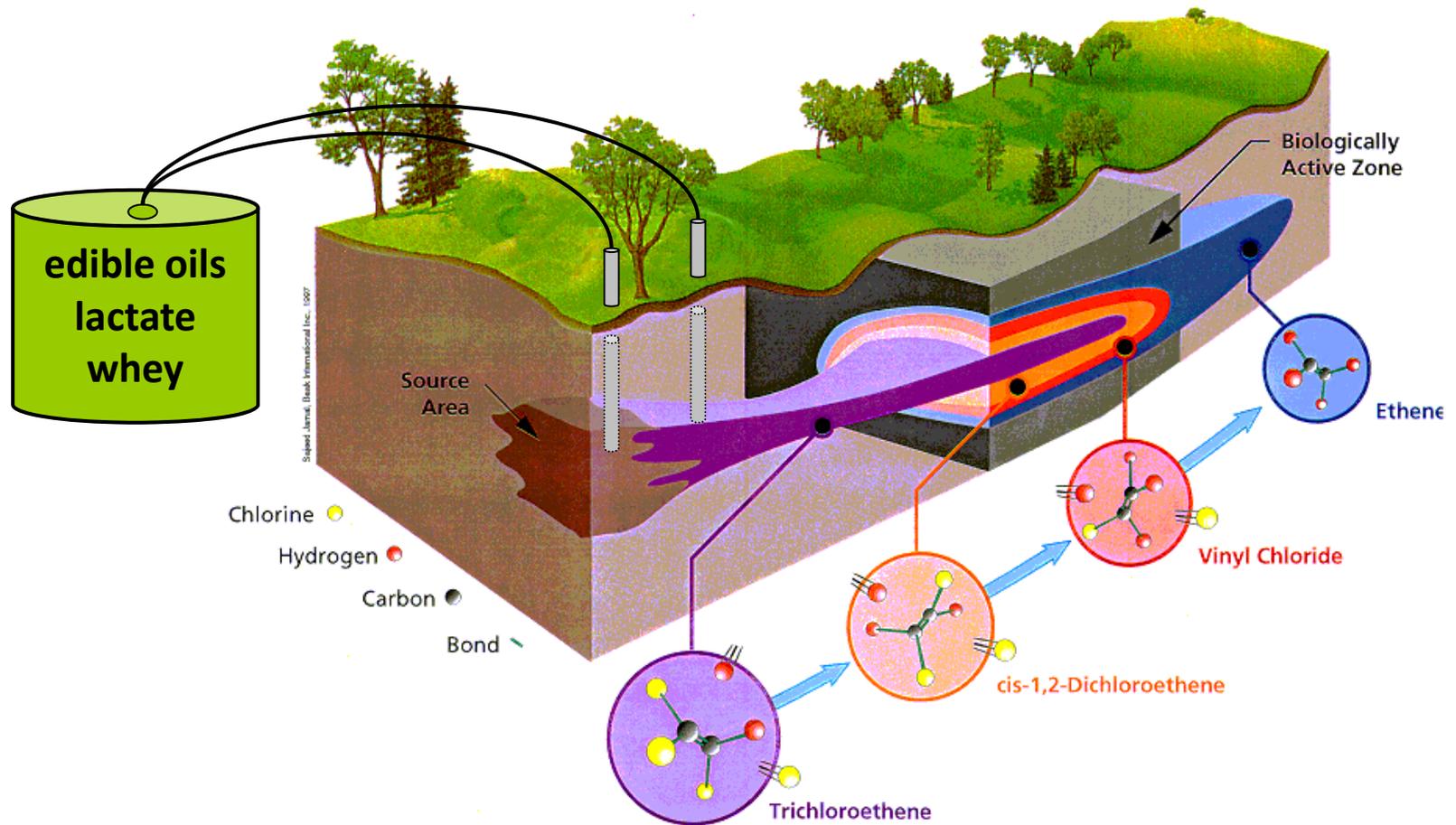


# Chloroethene reductive dehalogenases



# Their application in remediation

(enhanced) natural attenuation

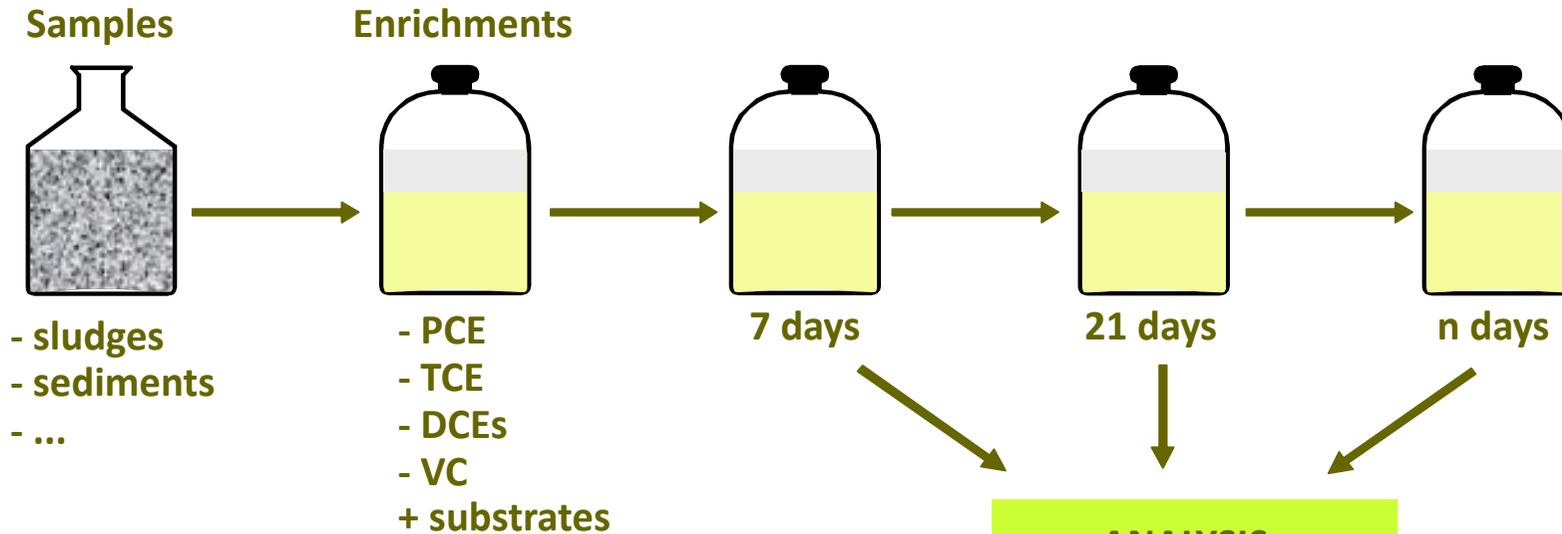


# Challenges for (enhanced) natural attenuation of chloroethene contamination

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- Assure that vinyl chloride is not end product
    - Is the biochemical potential present for complete chloroethene dechlorination ?
    - If yes, why does vinyl chloride accumulate ?
    - For enhanced NA which electron donors sustain best complete dechlorination ?
  
  - Understand the fate of toxic dechlorination products
    - Is dechlorination the only transformation process responsible for VC removal ?
-

# Experimental set-up of microcosms



**Different substrate mixtures added  
(1-2 mM per week)**

- acetate, formate, lactate, pyruvate, propionate, butyrate, ethanol
- acetate + formate
- propionate, butyrate, ethanol
- lactate + pyruvate

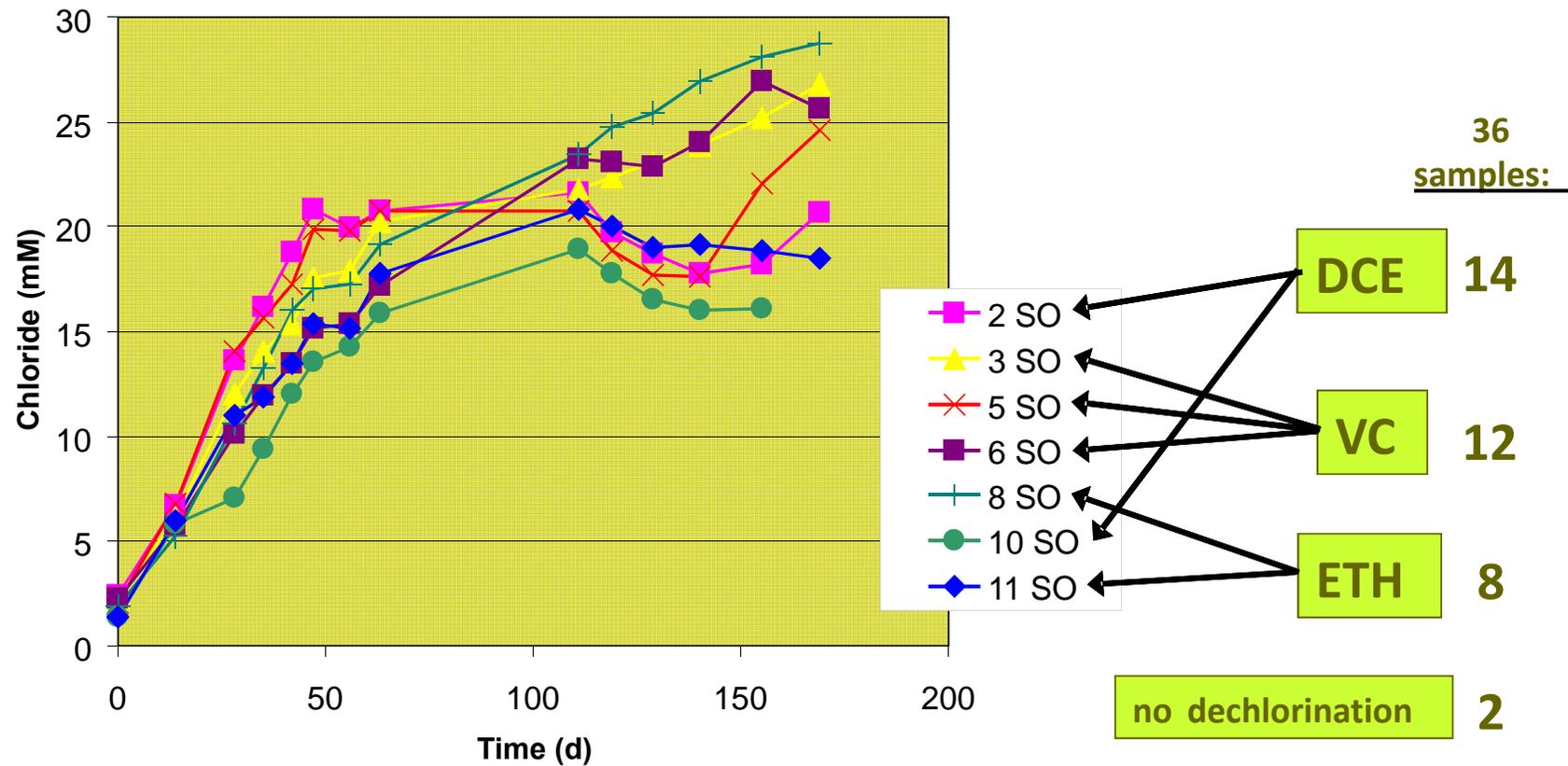
## ANALYSIS

- Chlor-o-counter
- GC
- ...

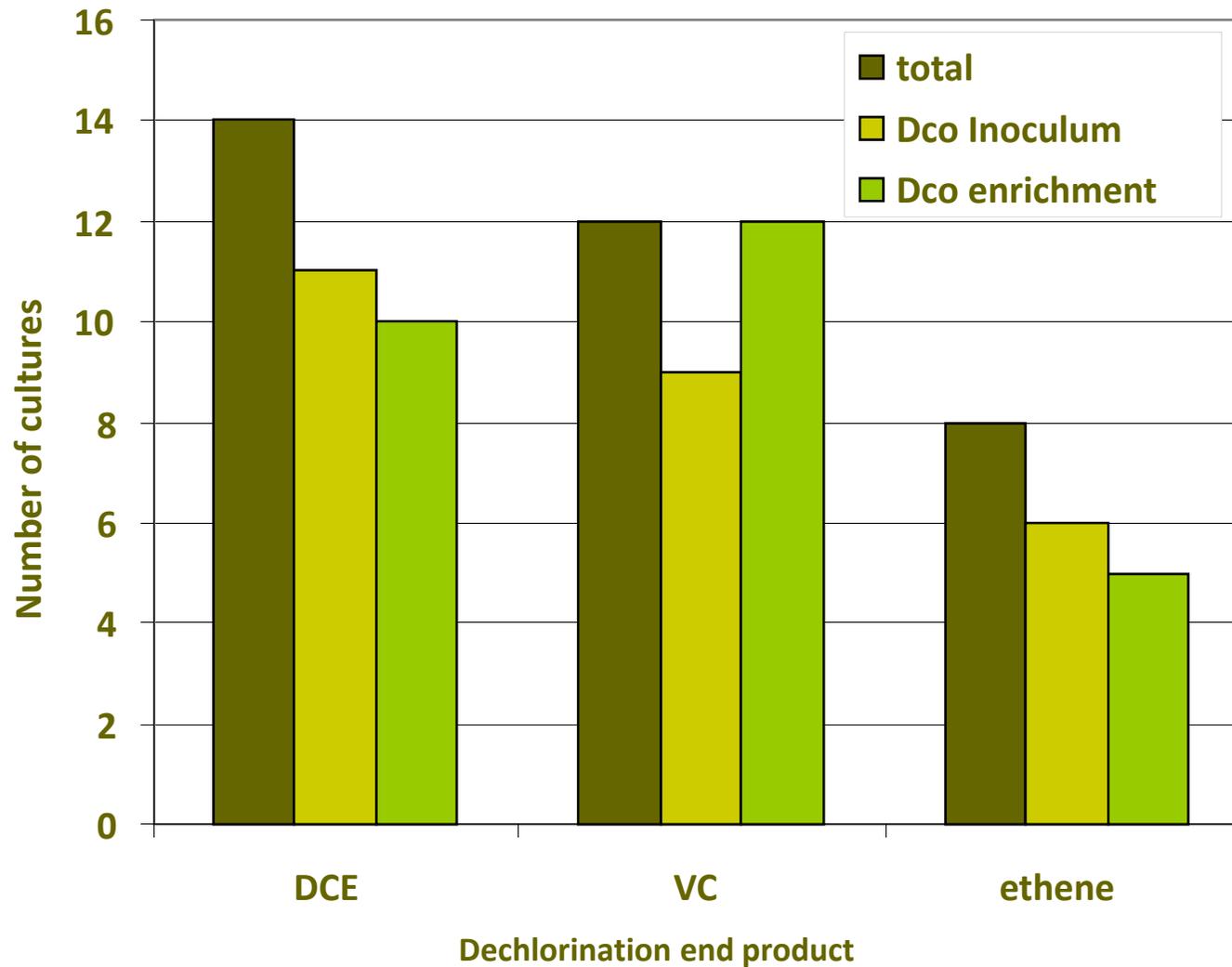
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**Isolation of DNA and  
RNA from samples  
and enrichments**

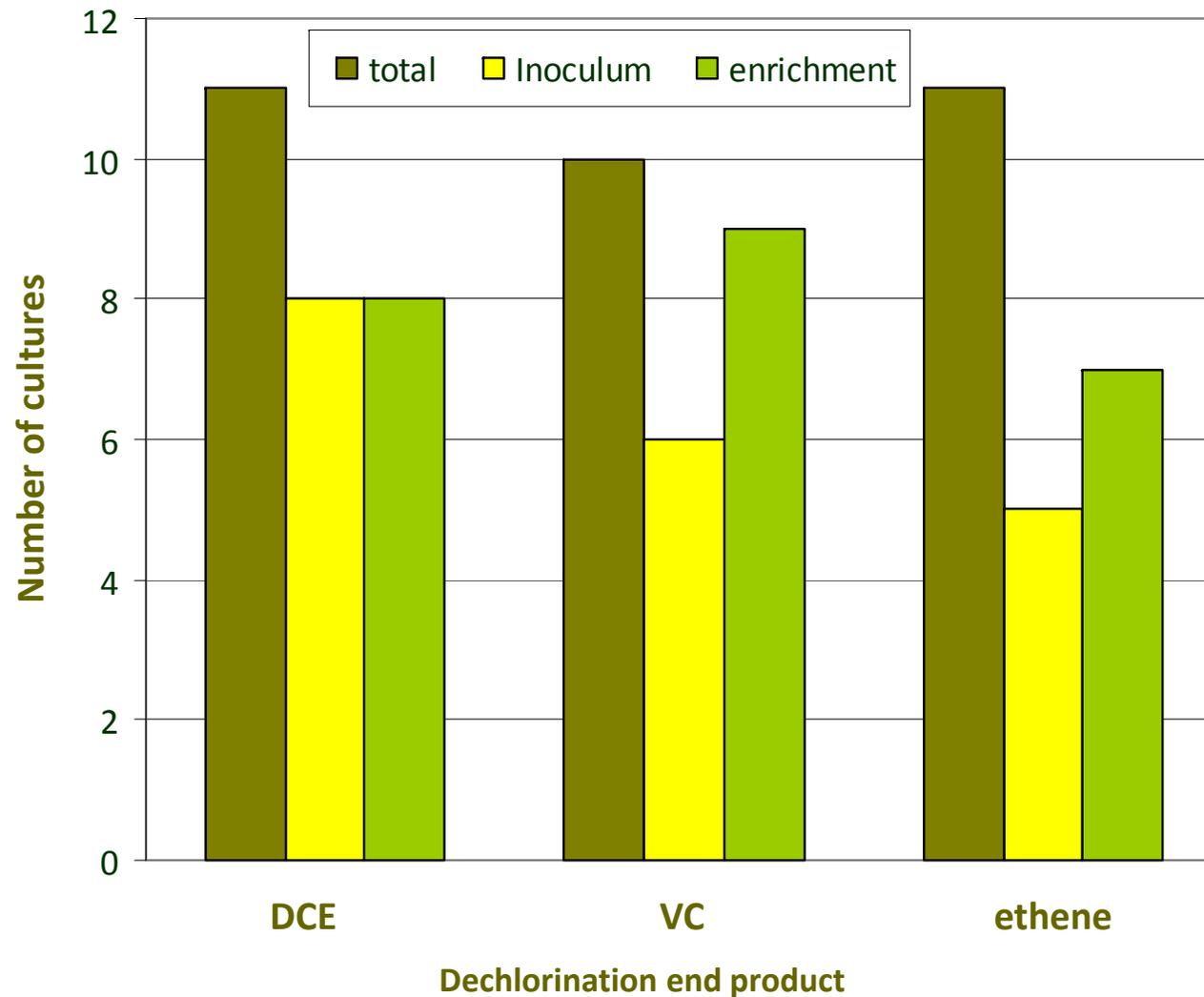
# Dechlorination in enrichment cultures



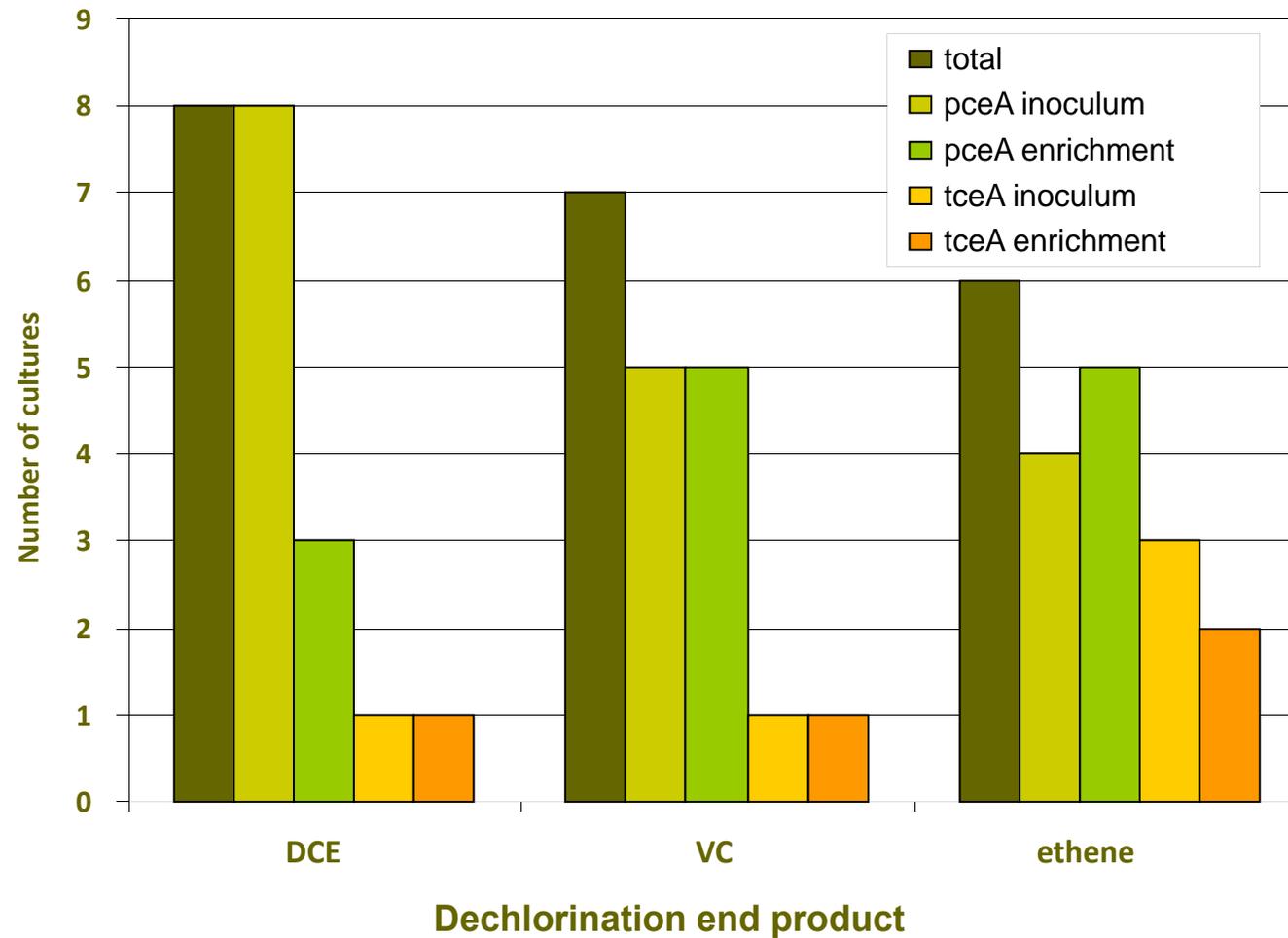
# Detection of *Dehalococcoides* spp. in 36 environmental samples



# Detection of *Dehalococcoides* spp. together with *Desulfitobacterium* spp. in 24/32 samples



# Detection of *pceA* et *tceA* in environmental samples



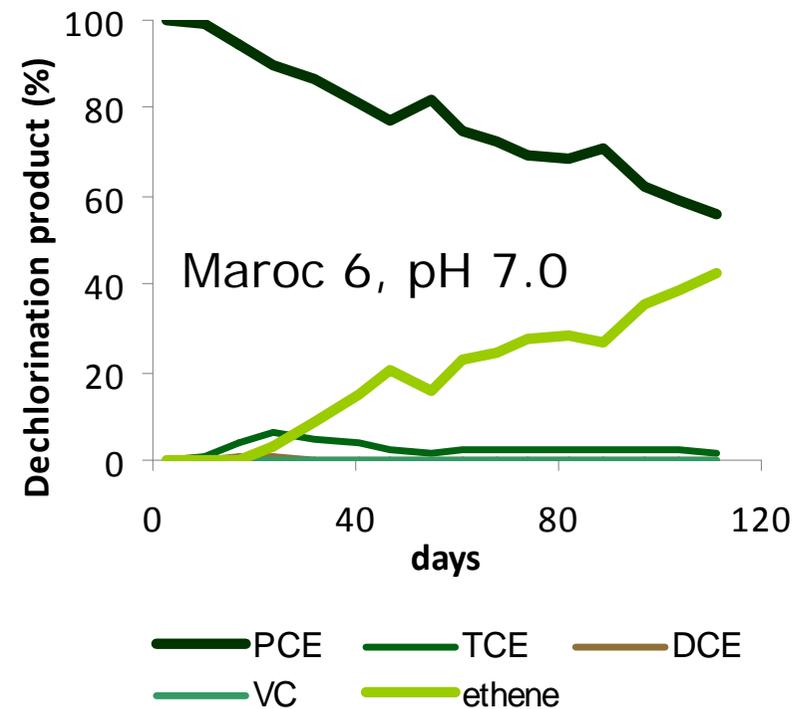
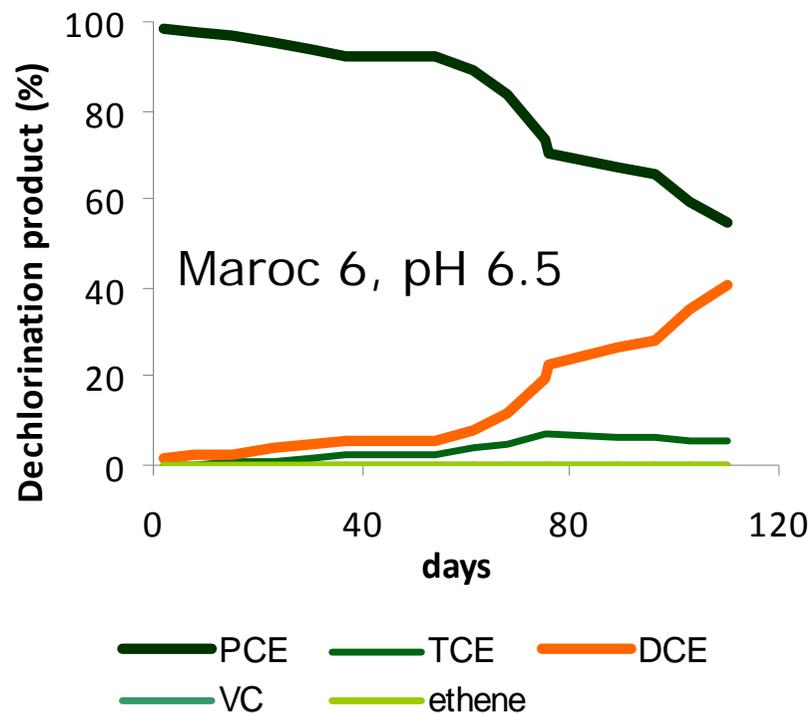
Inoculum: nested PCR; enrichment: direct PCR

# Stepwise dechlorination by different organohalide-respiring bacteria

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- PCE → DCE → ethene:
    - *Desulfitobacterium spp.* → *Dehalococcoides spp.*
  
  - PCE → DCE → ethene:
    - *Sulfurospirillum spp.* → *Dehalococcoides spp.*
  
  - PCE → DCE → ethene:
    - *Dehalobacter spp.* → *Dehalococcoides spp.*
    - **no vcrA and bvcA detected**
  
  - DCE → VC → ethene:
    - *Desulfitobacterium spp.* → *Dehalococcoides spp.*
    - **no vcrA and bvcA detected**
-

# The *Dehalobacter-Dehalococcoides* culture



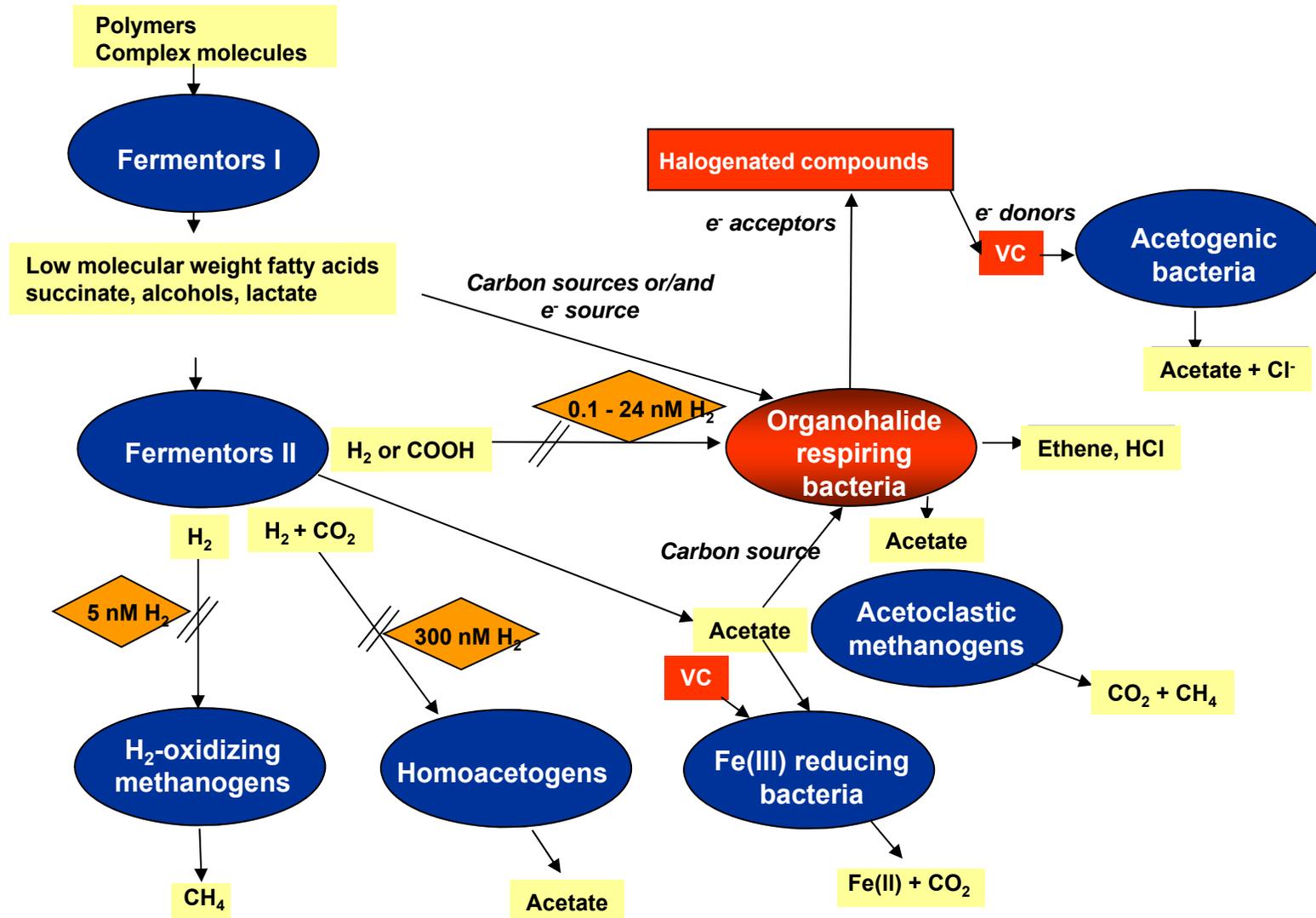
- very slow dechlorination
- production of DCE with low accumulation of TCE at pH 6.5
- production of ethene with low accumulation of TCE and no accumulation of DCE and VC at pH 7.0

# Challenges for (enhanced) natural attenuation of chloroethene contamination

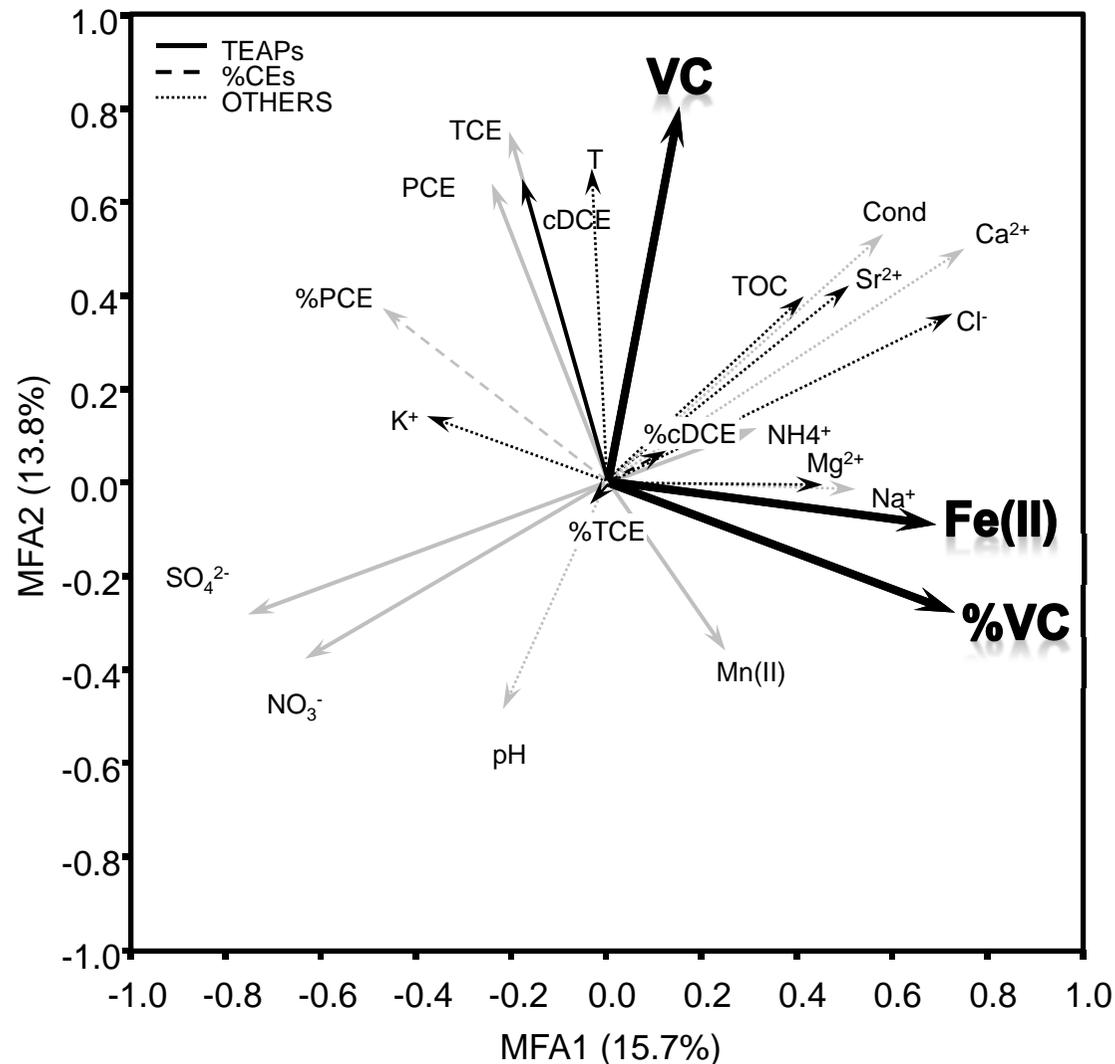
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# Bacterial guilds involved in organohalide respiration



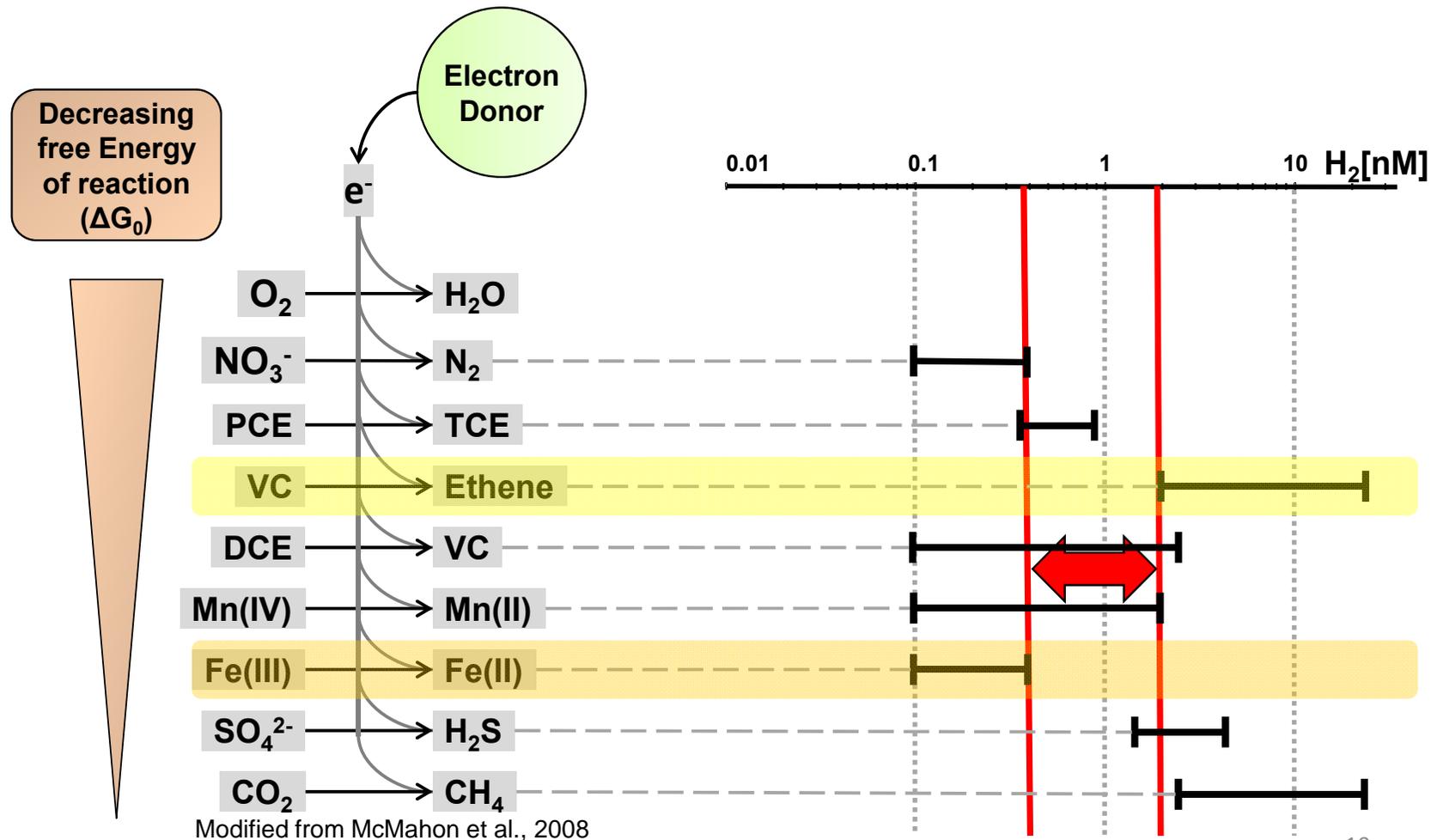
# Competition between iron-respiring and organohalide-respiring bacteria



# Competition between iron-respiring and organohalide-respiring bacteria

Ecological succession of electron acceptors

Required  $H_2$  concentrations



# Challenges for (enhanced) natural attenuation of chloroethene contamination

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# Electron donors for complete dechlorination

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Inoculum

Electron donor mixture leading to ethene formation

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Anaerobic digester sludge

Acetate - formate

PCE bioreactor sludge

Propionate – butyrate - ethanol

Aquifer I

Propionate – butyrate - ethanol

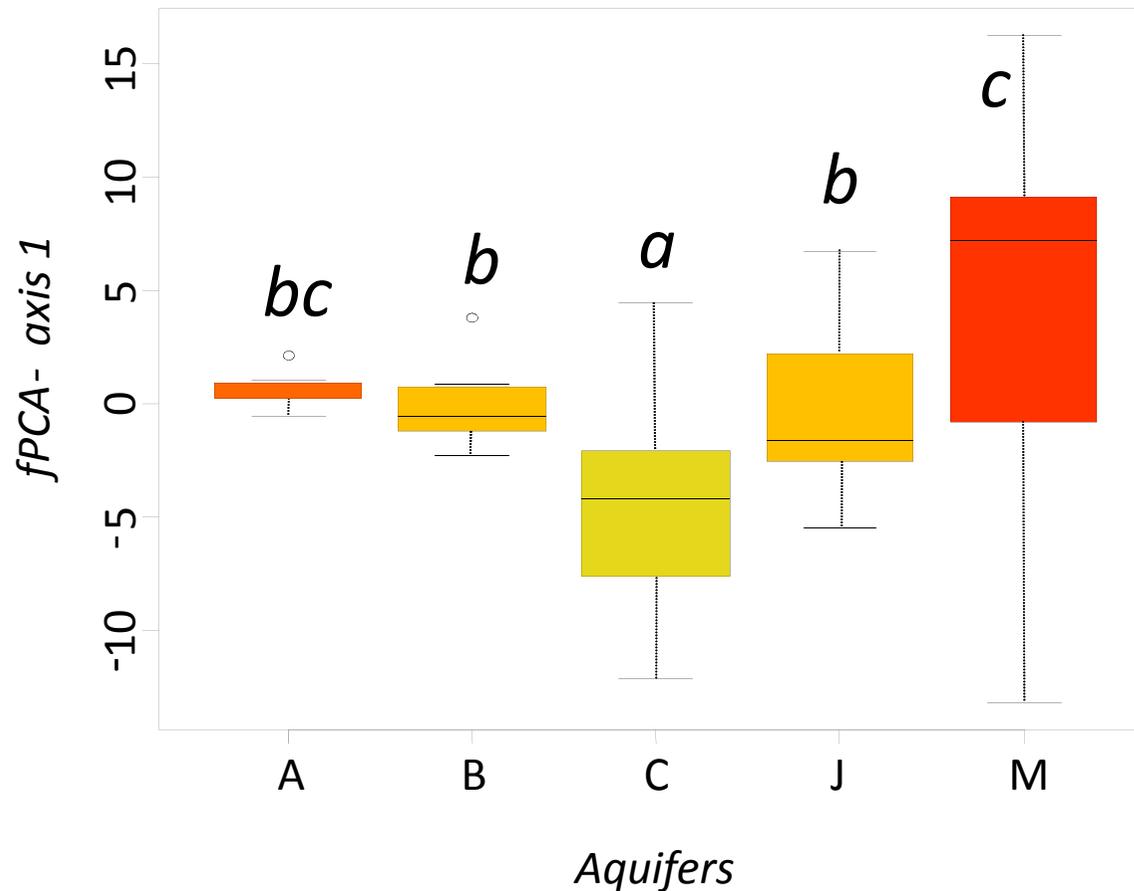
Aquifer II

Propionate – butyrate - ethanol

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# Bacterial diversity and microbial community composition varies according to aquifers



*Coordinates of record calculated on the first ordination axis of the fPCA, grouped by aquifers. Different letters indicate significant differences between the microbial communities of the five aquifers ( $P \leq 0.05$  Tukey's HSD).*

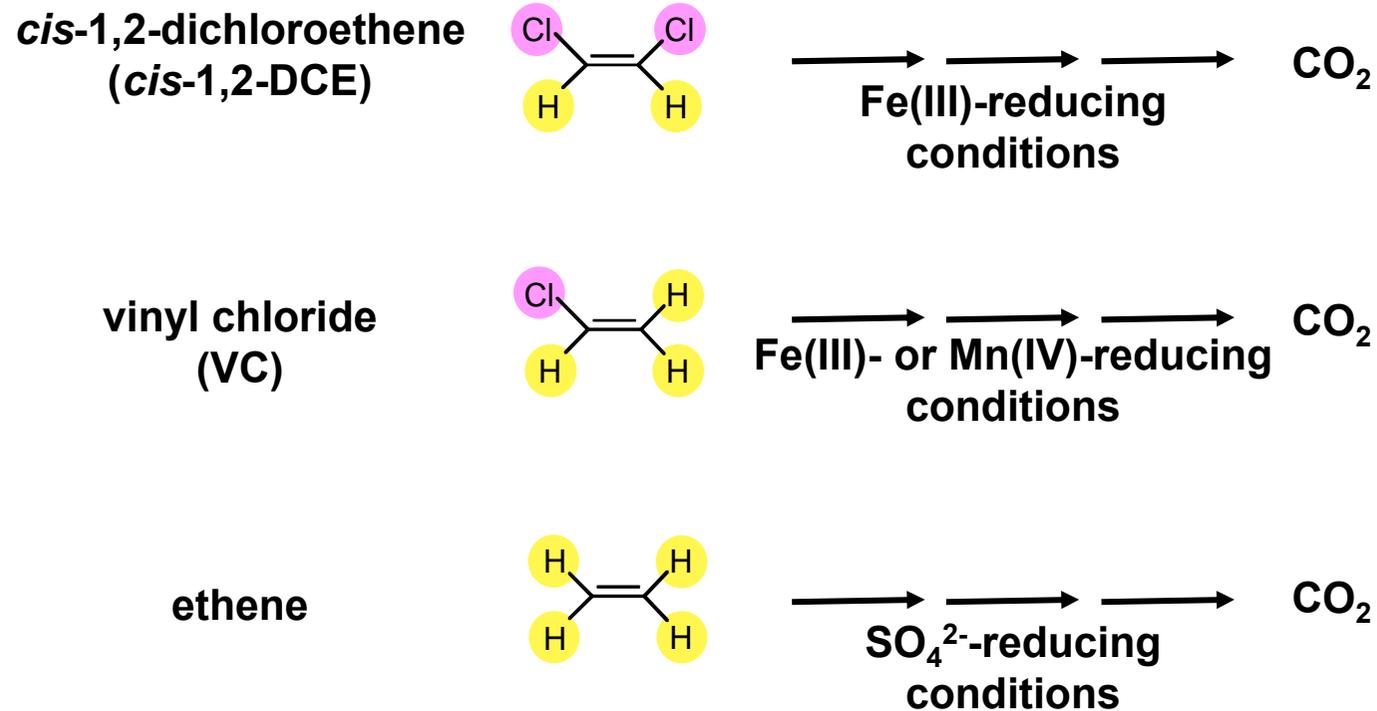
# Challenges for (enhanced) natural attenuation of chloroethene contamination

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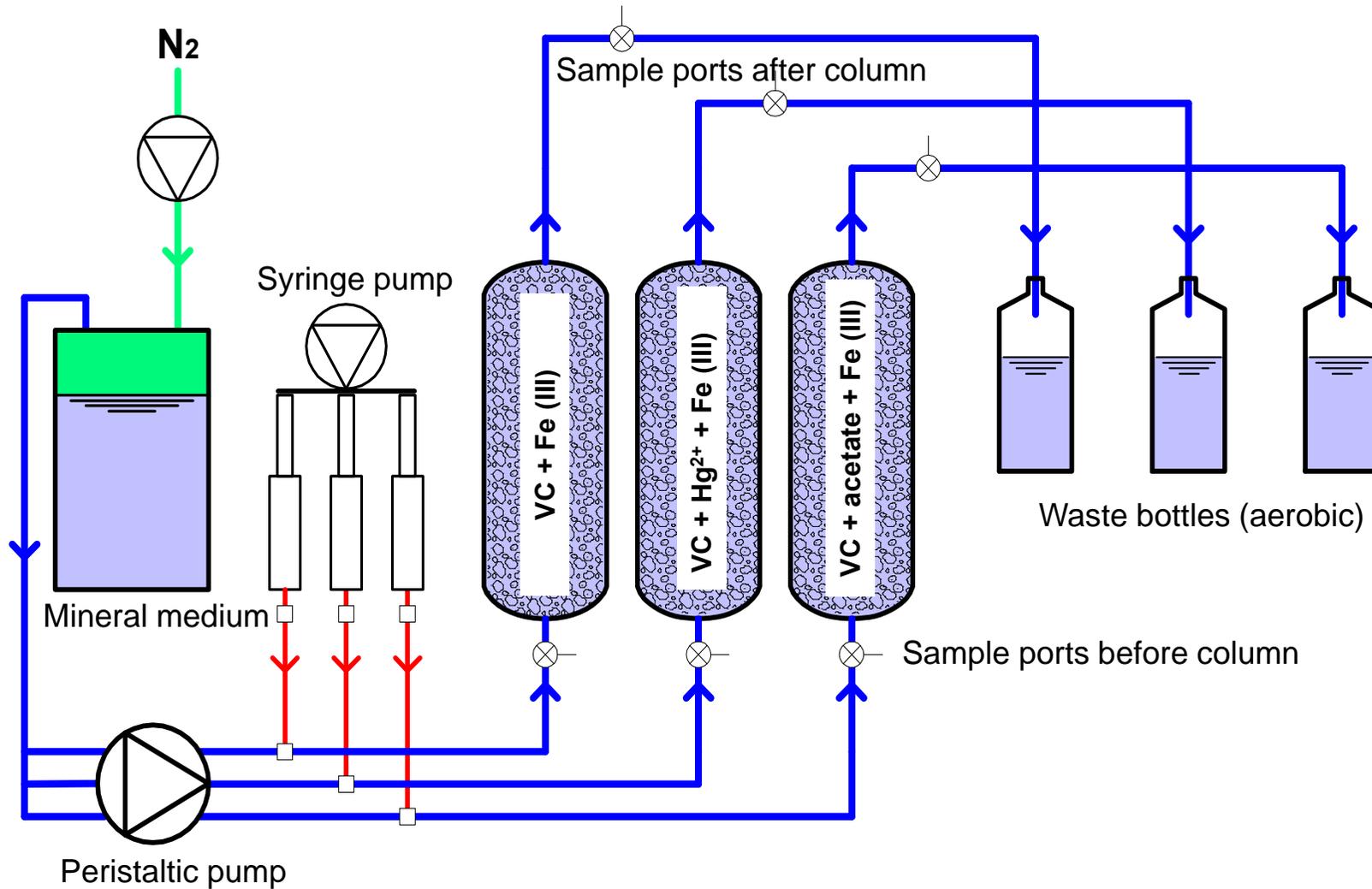
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# Anaerobic oxidation of chloroethenes

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# Scheme of experimental system



# Stable isotopes indicated anaerobic oxidation of VC

$$\delta^{13}\text{C}_{\text{out}} = \delta^{13}\text{C}_{\text{in}} + \epsilon \ln(\text{C}_{\text{out}}/\text{C}_{\text{in}})$$

$\epsilon$  = isotope enrichment factor

		$\epsilon$
→	VC + Fe (III)	→
	0-20% VC no ethene some Fe(II)	<hr style="width: 100%; border: 0.5px solid black; margin-bottom: 5px;"/> <b>- (3 – 5)</b>
→	VC + acetate + Fe (III)	→
	20-50% VC ethene Fe(II)	<b>- (11 – 18)*</b>

\* literature data for reductive dechlorination  $\epsilon = - (20-25)$

# Conclusions

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- Reductive dechlorination of chloroethenes is a promising microbial process for aquifer bioremediation
  - Organohalide-respiring bacteria seem to be widespread but not all biochemical potential seems to be everywhere
  - We do not know yet all the microbiology and enzymology involved in complete dechlorination to ethene
  - VC accumulation can be the result of missing biochemical potential or a consequence of losing competition for substrate
  - VC can also be transformed by processes other than reductive dechlorination
  - There is no general concept that can be applied to all sites, each site has its own characteristics and individual solutions have to be found
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# Final Conclusion

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- The bioremediation potential can be assessed with microbial investigations and a combination of molecular ecology, microcosms, and numerical ecology is the optimal approach to predict qualitatively the fate of chloroethenes during (enhanced) natural attenuation
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